

BONDING NOTES

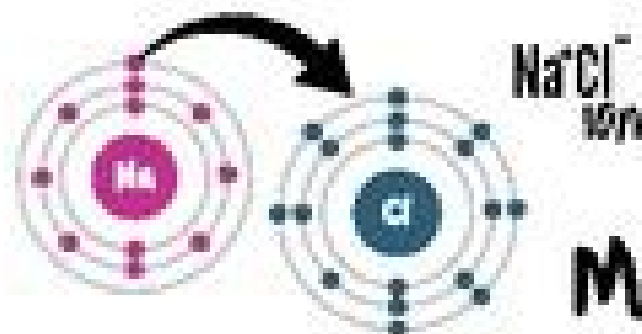
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IONIC BONDS

WHO IS BONDING: METAL • NON-METAL

HOW DOES IT WORK: METALS LOSE ELECTRONS & BECOME POSITIVELY CHARGED. POSITIVE IONS ARE CALLED "CATIONS."

NON-METALS GAIN ELECTRONS & BECOMES NEGATIVELY CHARGED. NEGATIVE IONS ARE CALLED "ANIONS."

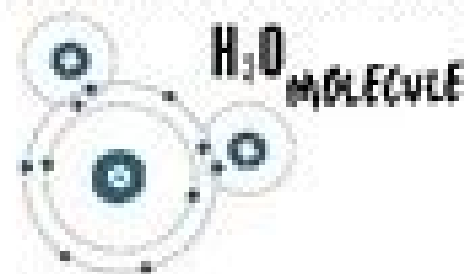


COVALENT BONDS

WHO IS BONDING: NON-METAL • NON-METAL

HOW DOES IT WORK: ELECTRONS ARE SHARED BETWEEN BOTH NONMETALS. THEY DO NOT TRANSFER.

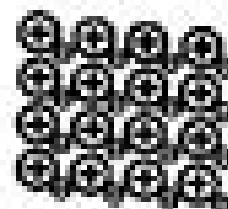
A MOLECULE IS POLAR IF THE ELECTRONS ARE UNEQUALLY SHARED. A MOLECULE IS NONPOLAR IF THE ELECTRONS ARE EQUALLY SHARED.



METALLIC BONDS

WHO IS BONDING: METAL • METAL

HOW DOES IT WORK: THE ELECTRONS FLOW FREELY BETWEEN THE POSITIVELY CHARGED PROTONS. THIS IS CALLED A SEA OF ELECTRONS. THIS CREATES A VERY STRONG BOND.



Sea of Electrons

Explorelearning Covalent Bonds Notes

Richard Bailey

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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