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Student Exploration: Nuclear Decay

Learning Objectives: After completing this activity, students should be able to:

- Describe the three types of nuclear decay: alpha, beta, and gamma.
- Explain how each type of decay affects the atomic number and mass number of a nucleus.
- Identify the particles emitted in each type of decay.

Prior Knowledge Questions (Must be answered before class):

1. What is a nucleus? (The central part of an atom, containing protons and neutrons.)

2. What is an alpha particle? (A helium nucleus, consisting of two protons and two neutrons.)

| Particle | Composition | Charge | Approximate Mass |
|----------|-----------------------|--------|------------------|
| Alpha | 2 protons, 2 neutrons | +2 | 4 amu |
| Beta | 1 electron | -1 | 1/1836 amu |
| Gamma | Photon | 0 | 0 amu |

3. How does the mass of a nucleus change after alpha decay? (It decreases by 4 units.)

4. How does the atomic number of a nucleus change after alpha decay? (It decreases by 2 units.)

5. How does the mass of a nucleus change after beta decay? (It remains the same.)

6. How does the atomic number of a nucleus change after beta decay? (It increases by 1 unit.)

Activity

1. In the simulation, select the "Alpha" button. Observe the alpha particle as it moves through the nucleus. What do you notice about its path? (It is a straight line.)

2. In the simulation, select the "Beta" button. Observe the beta particle as it moves through the nucleus. What do you notice about its path? (It is a wavy line.)



3. In the simulation, select the "Gamma" button. Observe the gamma ray as it moves through the nucleus. What do you notice about its path? (It is a straight line.)

4. In the simulation, select the "Alpha" button. Observe the alpha particle as it moves through the nucleus. What do you notice about its path? (It is a straight line.)

5. In the simulation, select the "Beta" button. Observe the beta particle as it moves through the nucleus. What do you notice about its path? (It is a wavy line.)

Explore Learning Nuclear Decay Answer

Tristan Sherwin



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Mathematics Learning Through Cooperative Small-Group Discovery Neil Davidson, James Fey, Charlene Beckmann, 2025-08-13 This book outlines cooperative small group discovery CSGD theory and practical learning strategies for implementing it in secondary and collegiate classrooms Based on Neil Davidson's decades of work the author team has designed a resource to help current users of small group methods in mathematics refine their practice and to entice others to try the strategies themselves The book describes principles and strategies for teaching complemented by an extensive

collection of examples from instructional materials designed to support teacher implementation with a focus on topics in the algebra curriculum Chapters are organized into four parts beginning with the theory and practice of CSGD and moving through examples and guidance both on sequencing CSGD activities into unit plans and addressing challenges of CSGD in the classroom The authors outline the rationale and basic operational principles of teaching through CSGD as well as common student and teacher roles accompanied by a variety of structural models to illustrate these roles The authors also include lesson plans that show how students can develop an understanding of elementary and advanced algebra through problem based CSGD and how coherent units of CSGD material can be used to develop student understanding of key ideas about linear and quadratic functions The authors complement this information with practical strategies for getting started with cooperative small group discovery teaching some common challenges in using small group methods and proven methods for solving those problems Ideal for educators and faculty involved in secondary and collegiate mathematics instruction this resource develops teacher understanding of principles and methods of cooperative learning and provides practical advice on getting started and refining that work

An Introduction to Chemistry Michael Mosher, Paul Kelter, 2023-03-18 This textbook is written to thoroughly cover the topic of introductory chemistry in detail with specific references to examples of topics in common or everyday life It provides a major overview of topics typically found in first year chemistry courses in the USA The textbook is written in a conversational question based format with a well defined problem solving strategy and presented in a way to encourage readers to think like a chemist and to think outside of the box Numerous examples are presented in every chapter to aid students and provide helpful self learning tools The topics are arranged throughout the textbook in a traditional approach to the subject with the primary audience being undergraduate students and advanced high school students of chemistry

Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order

Introductory Nuclear Physics Kenneth S. Krane, 1991-01-16 **INTRODUCTORY NUCLEAR PHYSICS** **Learning Electricity and Electronics**

with Advanced Educational Technology Michel Caillot, 2013-06-29 The objective of the NATO Advanced Research Workshop Learning electricity and electronics with advanced educational technology was to bring together researchers coming from different domains Electricity education is a domain where a lot of research has already been made The first meeting on electricity teaching was organized in 1984 by R Duit W Jung and C von Rhoneck in Ludwigsburg Germany Since then research has been going on and we can consider that the workshop was the successor of this first meeting Our goal was not to organize a workshop grouping only people producing software in the field of electricity education or more generally in the field of physics education even if this software was based on artificial intelligence techniques On the contrary we wanted this workshop to bring together researchers involved in the connection between cognitive science and the learning of a well defined domain such as electricity So during the workshop people doing research in physics education cognitive psychology and artificial intelligence had the opportunity to discuss and exchange These proceedings reflect the different points of view The main idea is that designing a learning environment needs the confrontation of different approaches The proceedings are organized in five parts which reflect these different aspects

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