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Student Exploration: Ionic Bonds

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

If Nate were to give Clara one of his markers, then both would end up with eight.

2. Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

If Maggie gives Fred and Florence one marker each, they will all have eight markers.

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The *Ionic Bonds* Gizmo allows you to explore how **ionic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (Cl)** are selected from the menus at right. Click **Play** () to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



1. Each atom consists of a central nucleus and several **shells** that contain **electrons**. The outermost electrons are called **valence electrons**.

How many valence electrons does each atom have? Sodium: 1 Chlorine:

7

2. Click **Pause** (). Elements can be classified as **metals** and **nonmetals**. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. **Electron affinity** is a measure of how tightly the valence electrons are held.

A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

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JE Gale



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Table of Contents Explore Learning Ionic Bonds Gizmo Answer Key

1. Understanding the eBook Explore Learning Ionic Bonds Gizmo Answer Key
 - The Rise of Digital Reading Explore Learning Ionic Bonds Gizmo Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Explore Learning Ionic Bonds Gizmo Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Explore Learning Ionic Bonds Gizmo Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Explore Learning Ionic Bonds Gizmo Answer Key
 - Personalized Recommendations
 - Explore Learning Ionic Bonds Gizmo Answer Key User Reviews and Ratings
 - Explore Learning Ionic Bonds Gizmo Answer Key and Bestseller Lists
5. Accessing Explore Learning Ionic Bonds Gizmo Answer Key Free and Paid eBooks
 - Explore Learning Ionic Bonds Gizmo Answer Key Public Domain eBooks
 - Explore Learning Ionic Bonds Gizmo Answer Key eBook Subscription Services
 - Explore Learning Ionic Bonds Gizmo Answer Key Budget-Friendly Options
6. Navigating Explore Learning Ionic Bonds Gizmo Answer Key eBook Formats

- ePub, PDF, MOBI, and More
- Explore Learning Ionic Bonds Gizmo Answer Key Compatibility with Devices
- Explore Learning Ionic Bonds Gizmo Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Explore Learning Ionic Bonds Gizmo Answer Key
 - Highlighting and Note-Taking Explore Learning Ionic Bonds Gizmo Answer Key
 - Interactive Elements Explore Learning Ionic Bonds Gizmo Answer Key
- 8. Staying Engaged with Explore Learning Ionic Bonds Gizmo Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Explore Learning Ionic Bonds Gizmo Answer Key
- 9. Balancing eBooks and Physical Books Explore Learning Ionic Bonds Gizmo Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Explore Learning Ionic Bonds Gizmo Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Explore Learning Ionic Bonds Gizmo Answer Key
 - Setting Reading Goals Explore Learning Ionic Bonds Gizmo Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Explore Learning Ionic Bonds Gizmo Answer Key
 - Fact-Checking eBook Content of Explore Learning Ionic Bonds Gizmo Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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