Balancing Chemical Equations – Answer Key

Balance the equations below:

- 1) 1 N₂ + 3 H₂ → 2 NH₃
- 2) 2 KClO3 → 2 KCl + 3 O2
- 3) 2 NaCl + 1 F2 → 2 NaF + 1 Cl2
- 4) 2 H₂ + 1 O₂ → 2 H₂O
- 5) 1 Pb(OH)₂ + 2 HCl → 2 H₂O + 1 PbCl₂
- 6) 2 AlBr₃ + 3 K₂SO₄ → 6 KBr + 1 Al₂(SO₄)₃
- 7) 1 CH4 + 2 O2 → 1 CO2 + 2 H2O
- 8) 1 C₃H₅ + 5 O₂ → 3 CO₂ + 4 H₂O
- 9) 2 CaH18 + 25 O2 → 16 CO2 + 18 H2O
- 10) 1 FeCl₃ + 3 NaOH → 1 Fe(OH)₃ + 3 NaCl
- 11) 4 P + 5 O2→ 2 P2O5
- 12) 2 Na + 2 H₂O → 2 NaOH + 1 H₂
- 13) 2 Ag₂O → 4 Ag + 1 O₂
- 14) 1 Se + 12 O2 → 8 SOs
- 15) 6 CO2 + 6 H2O → 1 C6H12O6 + 6 O2
- 16) 1 K + 1 MgBr → 1 KBr + 1 Mg
- 17) 2 HCl + 1 CaCO₃ → 1 CaCl₂ + 1 H₂O + 1 CO₂
- 18) 1 HNO3 + 1 NaHCO3 → 1 NaNO3 + 1 H2O + 1 CO2
- 19) 2 H₂O + 1 O₂ → 2 H₂O₂
- 20) 2 NaBr + 1 CaF2 → 2 NaF + 1 CaBr2
- 21) 1 H₂SO4 + 2 NaNO2 → 2 HNO2 + 1 Na2SO4

Explore Learning Balancing Chemical Equations Answer Key

Lauren Gardner

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