### AICE Marine Science Semester 1 Review

#### Unit 1 - Water

#### Key Vocabulary:

Kinetic particle theory
 Water vapour
 Proton
 Neutron
 Electron
 Nucleus
 Shells
 Atomic number
 Bond
 Emergent properties
 Covalent bond
 Ionic bond

Hydrogen bond

Polar
Solvent
Density
Specific heat capacity
Solute
Solution
Solubility
Dissolution
Salinity
Precipitation
Run-off
Evaporation
Hypersaline
Halocline

Thermocline
PH
Acidic
Alkaline
Neutral
PH scale
Turbulence
Dissolved oxygen
Phytoplankton
Oxygen minimum layer
Water column
Gradient
Thermal insulator
Pycnocline

- 1. List the 5 oceans and know their locations
- Identify the 4 main layers/zones of the ocean and which ones are photic?
- 3. Identify when you would use the different words: pelagic and benthic?
- 4. Explain the kinetic particle theory of WATER and the differences between solids, liquids and gases.
- 5. Identify the following items found in the Ocean
  - a. Key elements
  - Key compounds (ionic and covalent)
  - c. Key mixtures
- Draw the structure of the following atoms (include nucleus, protons, neutrons, and electrons (show arrangement of electrons and shells/rings))
  - a. Oxygen
  - b. Hydrogen
  - c. Carbon
  - d. Sodium
  - e. Chlorine-35
- Using a diagram describe the covalent bonding in a water molecule including the sharing of electron pairs.

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**Sebastian Brünink** 

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