

Skills Worksheet

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

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U.S. Atomic Energy Commission



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Holt Chemistry Ralph Thomas Myers, 2004 **Holt Chemistry** Holt Rinehart & Winston, 2003-01-24 **Merrill Chemistry** Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **National Library of Medicine Current Catalog** National Library of Medicine (U.S.), **Organic Chemistry** T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2023

Organic Chemistry 13th edition provides a comprehensive yet accessible treatment of all the essential organic chemistry concepts with emphasis on relationship between structure and reactivity in the subject. The textbook includes all the concepts covered in a typical organic chemistry textbook but is unique in its skill development approach to the subject. Numerous hands-on activities and real-world examples are integrated throughout the text to help students understand both the why and the how behind organic chemistry. This International Adaptation offers new and updated content with improved presentation of all course material. It offers new material on several topics including the relevance of intermolecular forces in the immune response and vaccines like those for Covid 19, the chemistry of breathing carbonic anhydrase, how conjugation and complexation affect the color of lobsters, and how biodegradable polymers are used to stabilize vaccines and pharmaceuticals. Content is revised to reflect the current understanding of chemical processes and improved depictions of longstanding mechanisms. This edition builds on the ongoing pedagogical strength of the book with the inclusion of additional worked and end-of-chapter problems and an engaging set of new problems entitled Chemical Consultant Needed. These draw from the primary chemical literature and give students experience of working with more complex polyfunctional structures and areas where key transformations take place.

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