

Single Loop Bracelet Instructions

You Will Need:
28 Bands &
1 S-Clip

1. Start with the Tab end of the Loom facing you and refer to the Place Pattern. Place a Band on the Loom around Posts 2 & 3. Next place a Band around Posts 3 & 5, then Posts 5 & 6, and so on (figure 7). Follow this pattern all the way along the Loom, ending at Post 42. Your loom should look like figure 8.



figure 7

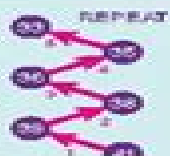


figure 8

Single Loop Bracelet PLACE PATTERN



Single Loop Bracelet PICK PATTERN



2. Rotate the Loom so that the Notch end faces you. Refer to the Pick Pattern.
3. Find the Band that connects Posts 39 & 41 (figure 9). Pick the end of that Band that is around Post 41 and loop it over Post 39 (figure 10).
4. Find the Band that connects Posts 38 & 39 (figure 11). Pick the end of that Band that is around Post 39 and loop it over Post 38 (figure 12). Note that the Band you need to pick is below the other looped Bands.



figure 9



figure 10



figure 11



figure 12



figure 13

5. Continue picking and looping in this zig-zag pattern until you've looped a Band onto Post 2. Your Loom should look like figure 13.
6. Follow the instructions in the "Finishing It Off" section to finish your Single Loop Bracelet.

Your set will make up to 24 bracelets if you use 24 Bands and end on Post Number 38 instead of continuing to Peg 42. Once all Bands have been placed, find the Band that connects Posts 35 & 36. Pick the end of that Band that is around Post 36 and loop it over Post 35. Find the Band that connects Posts 33 & 35. Pick the end of that Band that is around Post 35 and loop it over Post 33. Continue picking and looping in this zig-zag pattern until you've looped a Band onto Post 2. Follow the instructions in the "Finishing It Off" section to finish your Single Loop Bracelet. The result will be a slightly snugger bracelet.



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United States. SIC Coding Task Group

A red circular graphic with a gradient, appearing as a partial circle or a thick arrow pointing to the right, located to the right of the text 'United States. SIC Coding Task Group'.

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novices and experienced knitters featuring clear instructions illustrations and a variety of patterns that reflect the 19th century revival of craft based skills Cupples literary style is infused with a nurturing tone encouraging creativity while preserving the intricate traditions of knitting The contextual backdrop of the Victorian era further enriches the text as domestic crafts were not only a pastime but also a social necessity illuminating the intersection of gender labor and artistry George Mrs Cupples emerged from a rich tapestry of literary and crafting influences embodying the spirit of the Women s Movement during her time This influential author combined her passion for knitting and pedagogical dedication likely motivated by a desire to empower women through the preservation of handicrafts Her text serves as both a practical guide and a cultural artifact reflecting the domestic realities and aspirations of her era For anyone interested in the knitting tradition or seeking to enhance their craft The Stocking Knitter s Manual is an invaluable resource This book not only imparts technical skills but also celebrates the artistry behind each knitted piece It is a must read for both historians and crafters ensuring lasting inspiration for generations to come The Stocking-knitter's Manual. A Handy Book for the Work-table Mrs. George Cupples,1868 The stocking-knitter's manual Anne Jane Cupples,1870 Patternalia Jude Stewart,2015-10-13 From the author and designer of ROY G BIV a delightful fully illustrated new volume on patterns from polka dots to plaid their histories cultural resonances and hidden meanings Art and Industry: (1892) Industrial and manual training in the public schools United States. Office of Education,Isaac Edwards Clarke,1892 Choice ,1977

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is $2\text{Cl}^- + 2\text{Ag}^+$

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver i ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13 2ag aq 2f aq 2nh₄ aq cr₂o₇²⁻ aq ag₂cr₂o₇ s 2nh₄ aq 2f aq because two nh₄ aq

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