

## III. CHEMISTRY NETWORK LENSES

### CHEMICAL REACTIONS

#### GRADE 10-11

1. Hydrogen gas is produced with large quantities of carbon monoxide. The hydrogen then undergoes a series of chemical reactions (process 1) and carbon monoxide enters the water-gas shift and the water-gas reaction (process 2).

	Water-gas reaction	Water-gas shift
A	Not modeled	Not modeled
B	Not modeled	Not modeled
C	Not modeled	Not modeled
D	Not modeled	Not modeled

2. Feed streams to unit 10, water-gas shift, are summarized in the following equation:

$$10 \text{ H}_2 + 10 \text{ CO} \rightarrow 10 \text{ H}_2\text{O}$$

The water-gas shift reaction is summarized as:



After  
shift

Feed to the water-gas shift reaction is summarized as:

	CO	H <sub>2</sub>
A	1	1
B	1	1
C	1	1
D	1	1

3. The water-gas shift and water-gas reactions are summarized in the following table for a particular

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