

Chemistry – Unit 4 Worksheet 4

DMC.5: *I can cite and interpret the evidence that supports the belief that some pure substances are made up of simpler particles (atoms) in a definite ratio.*

Answer the following questions. Be sure to show all mathematical work and reasoning and use complete sentences in explanations.

1. Two compounds of hydrogen and oxygen are tested. Compound I contains 120.0 g of oxygen and 15.0 g of hydrogen. Compound II contains 32.0 g of oxygen and 2.0 g of hydrogen.
 - a. Determine the ratio of the mass of oxygen to the mass of hydrogen in each of the compounds.
 - b. Why are the compounds not the same?
 - c. What is significant about these mass ratios?
 - d. If compound I is water, what could be the formula of compound II?
2. Nitrogen and oxygen combine to form a variety of compounds. The following data were collected for three different compounds of nitrogen and oxygen:

Analysis Data of Nitrogen & Oxygen Compounds	
Compound	Mass of Nitrogen that combines with every 1.0 g of Oxygen
A	1.750 g
B	0.8750 g
C	0.4375 g

- a. Additional evidence shows that the formula of compound B is NO. Sketch particle diagrams of molecules of all three compounds.
- b. Justify your representations above.
- c. There is no compound between Nitrogen and oxygen with a mass ratio of 2.00 grams of N for every 1.00 grams of oxygen. Why?

Modeling Chemistry Utest Answers

M. L. Kahn



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Model Questions and Answers in Chemistry Marthinus Jacobus Oosthuizen,1970 **Model Answers on Chemistry** Elizabeth Joan Perkins,1967 **Model Questions and Answers in Chemistry** Marthinus Jacobus Oosthuizen,1972

Chemistry (Advanced Level) Louis Alexander Linton,1965 **Chemistry Model Answers** RND Publications Ltd,1998-12-01 **Model Answers in Organic Chemistry** A. J. Showler,T. A. Brown,2016-06-06 **Model Answers in Organic Chemistry** **Chemistry (Advanced Level)** Louis Alexander Linton,1960 Model Answers M. A. Phillips,1958

Advanced Level Chemistry Vijay Pandit,1990-04-01 **Model Answers G.C.E. Advanced Level Chemistry** Vijay Pandit,1981 **Chemistry for Senior Students** Charles Alfred Janson,1986 **Model Answers on Chemistry. (General Certificate of Education.)**. Elizabeth Joan Perkins,1967 *Model Answers* ,1988 **Model Answers: Chemistry Gcse** RND Publications Ltd, Pure Chemistry ,1975 **Chemistry Equations and Answers** Mark Jackson,2006-02-22 This 6 page study guide contains basic chemistry analysis and concepts designed specifically to aid science students *Model Answers* M. L. Kahn,1989 Model Answers G. Georgiou,1995 New Rudman's Questions and Answers on the Graduate Record Advanced Test in Chemistry Jack Rudman,National Learning Corporation,1974 *Model Answers* M. L. Kahn,1989-01-01

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