

Chemistry – Unit 6 Worksheet 1

After having observed molecular models which show 3D shape comparisons were discussed. After creating particle level models for binary compounds charged ions, bond length, weight, mass, molar mass, particle level model for binary compounds charged ions (ammonia), hydrogen, sodium chloride, water, methyl alcohol, particle level model, water, ammonia, the three compounds listed are used to illustrate effects of size and weight. After each model is created the students must think about the relative size of the atoms and the relative mass of the atoms to predict the differences in the charges of the ions and the shape of the molecules.

1. After the model is created, the students must create a particle level model of the compound. The name and formula of the compound is given.

Binary compound formula	1. ammonia 2. nitrogen	2. hydrogen 3. nitrogen	3. ammonia 4. sulfur	4. hydrogen 5. sulfur
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particle level model



Binary compound formula	5. ammonia 6. nitrogen	6. hydrogen 7. nitrogen	7. ammonia 8. sulfur
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particle level model



Binary compound formula	8. ammonia 9. nitrogen	9. hydrogen 10. nitrogen	10. ammonia 11. sulfur	11. ammonia 12. nitrogen
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particle level model



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