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## Chemistry - Unit 7 Review

#### Chemical Reaction Model

Describe key characteristics of all chemical reactions, including the rule of energy.
Explain how a balanced equation represents these features unclude an enough).

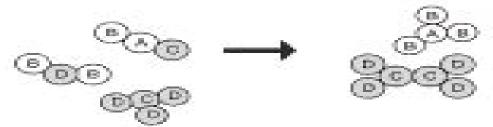
In chemical exactions, atoms of the reactants recombine to from new substances in the products. There is almost always some transfer between the thermal and chemical energy accounts as well as an exchange of thermal energy via heating 600 between the system and sucroundings. The bolimaced chemical equation represents this process symbolically, showing that the atoms in the products are the same as these in the reactions.

2 H<sub>s</sub> + O<sub>v</sub> → 2 H<sub>s</sub>O + energy

Note: there are 4 atoms of H and 2 atoms of O an each side of the equation.

#### Applying the Model

Balance the reaction below by drawing any additional reaction and/or product molecules that would be needed.



Write the balanced equation for this reaction.

$$DB_2 + CD_3 + BAC \rightarrow C_2D_4 + AB_6$$

Which of these nulecules would be called reactants?

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