

Chemistry – Unit 6 Worksheet 1

We have observed evidence that atoms of different elements are different, the most noticeable being their predictable characteristic chemical behaviour. Atoms of one element react in a predictable way with atoms of another element. However, when the elements react, sometimes several particles are combined to form compounds. They do not combine randomly as solids. We will now examine the patterns that exist for the ratios in which these elements combine to form the compounds that they do.

1. Write the formula and draw the particle diagram for each compound. The ratio of atoms in each compound is given.

Atomic formula ratio	12 hydrogen 16 oxygen	12 hydrogen 16 oxygen	12 hydrogen 16 oxygen	1 hydrogen 1 oxygen
	1:2:2	1:2:2	1:2:2	1:2

particle



diagram

Atomic formula ratio	12 hydrogen 16 oxygen	12 hydrogen 16 oxygen	12 hydrogen 16 oxygen
	1:2:2	1:2:2	1:2:2

particle



diagram

Atomic formula ratio	12 hydrogen 16 oxygen	12 hydrogen 16 oxygen	1 hydrogen 16 oxygen	1 hydrogen 16 oxygen
	1:2:2	1:2:2	1:2:2	1:2:2

particle



diagram

Modeling Chemistry Unit 6 Test Review

Christian Drosten



Modeling Chemistry Unit 6 Test Review:

Decoding **Modeling Chemistry Unit 6 Test Review**: Revealing the Captivating Potential of Verbal Expression

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