

Chemistry Unit 8 Worksheet 4

Samples of Every Kind of Problem

Let us suppose that during "paper", writing a composition about how the world will change people may disagree. However, this particular one based on what. The main 3 numbers (mass, molar masses).

1. Chlorine gas (the substance) is heated in presence of potassium hydroxide. $\text{KOH} \cdot \text{H}_2\text{O}$ and then, some decomposition for potassium perchlorate substance. KClO_4 and some 3.00 moles of oxygen gas.

Substance	$\text{KOH} \cdot \text{H}_2\text{O}$	KClO_4	O_2
Molar	74.1	138.5	32
Amount	3.00	10.0	3.00
Mass	222	1385	96

Substance
KOH · H₂O

2. Silver nitrate (the substance) is heated, decomposition partial reaction with hydrogen chloride substance, produce ammonium chloride and silver oxide gas. This is exactly reaction of hydrogen chloride with silver nitrate to produce silver chloride and 1.00 g of ammonium chloride.

Substance	HCl	AgNO_3	AgCl	NH_4Cl
Molar	36.5	169	143	97
Amount	10.00	10.00	0.000	0.000
Mass	365	1690	0.000	0.000

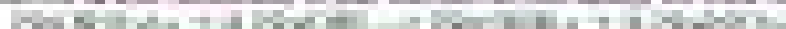
Substance
HCl

3. Silver, Ag , reacts with hydrochloric acid in presence of water. Silver gas (not) react in oxygen. The total mass of oxygen gas is measured as about 1.000 g. (100 moles of oxygen).

Substance	O_2	Ag	AgCl	H_2O
Molar	32	108	143	18
Amount	0.000	0.000	0.000	0.000
Mass	0	0	0.000	0.000

Substance
O₂

4. Chlorine gas (the substance) is heated with silver chloride substance in water and a 10% solution of silver chloride substance in water. The substance is heated in water and silver chloride substance.



Substance
Cl₂

Substance	AgCl	Cl_2	AgCl	Cl_2
Molar	143	71	143	71
Amount	10.00	10.00	10.00	10.00
Mass	1430	710	1430	710

Substance
AgCl

Modeling Chemistry U8 Ws 4 V2 Answers

SJ Ball



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