

Chemistry Unit 8 Worksheet 8

Examples of Every Kind of Problem

This is a separate sheet of paper, so this is a complete solution to each of the problems. You can follow the problems without an answer. We want to make your final solution.

1. Calculate the reaction of sulfur of potassium chlorate, KClO_3 , and potassium chlorate to produce potassium chloride KCl and O_2 reaction of oxygen gas.



Mass:	1.0	0	0
Volume:	1.0	1.0	1.0
Mass:	0	1.0	1.0

1.0 g of KClO_3 is needed

2. In a single displacement reaction, potassium metal reacts with hydrochloric acid to produce potassium chloride and hydrogen gas. How many moles of hydrochloric acid are needed to completely react with 1.0 g of potassium?



Mass:	1.0	0	0	0
Volume:	1.0	1.0	1.0	1.0
Mass:	0	0	1.0	1.0

1.0 g of KCl is needed

3. Ethanol, $\text{C}_2\text{H}_5\text{OH}$, reacts with oxygen gas to produce carbon dioxide gas and water vapor. What mass of oxygen gas is required to react with 1.0 g of ethanol?



Mass:	1.0	0	0	0
Volume:	1.0	1.0	1.0	1.0
Mass:	0	0	1.0	1.0

1.0 g of $\text{C}_2\text{H}_5\text{OH}$ is needed. 1.0 g of O_2 is needed.

4. Sodium metal reacts with sulfuric acid to produce sodium sulfate and hydrogen gas. How many moles of sulfuric acid are needed to react with 1.0 g of sodium?



Mass:	1.0	0	0	0
Volume:	1.0	1.0	1.0	1.0
Mass:	0	0	1.0	1.0

1.0 g of Na_2SO_4 is needed. 1.0 g of H_2SO_4 is needed.

Modeling Chemistry U8 Ws 3 V2 Key

Rachel S Tattersall



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