

## Unit 7 Chemistry - Nail Lab

**Purpose:** After you have completed the lab, think about what was the overall goal for performing the lab. What were we trying to accomplish?

### Procedure:

#### Step 1

1. Mass a clean & labeled \_\_\_\_\_ mL beaker.
2. Put between \_\_\_\_\_ g of copper (II) chloride in the beaker. Record the mass of the beaker and the copper (II) chloride.
3. Add \_\_\_\_\_ mL of distilled water to the beaker. Stir to dissolve the solid.
4. Obtain the mass of \_\_\_\_\_ iron nails together to ~ 0.5 g.
5. Place the nails in the copper (II) chloride solution. Observe the reactions; record your observations. Place the labeled beaker in the place designated by your teacher.

#### Step 2

- a. Remove the nails. Rinse or scrape the \_\_\_\_\_ (copper metal) from the nails into your labeled beaker. Place the nails in a separate labeled small beaker. Note the appearance of the nails in your observations. Place this beaker in the drying oven.
- b. \_\_\_\_\_ the solution from the 250 mL beaker. Rinse the precipitate with about \_\_\_\_\_ mL of distilled water. Try to lose as little of the solid copper as you can when you decant. Add a 2<sup>nd</sup> rinse with distilled water, rinse the copper with \_\_\_\_\_ . Rinse your last time with distilled water. Then place the labeled beaker in the drying oven.
- c. Mass the dry nails, and then discard the nails.

#### Step 3

- a. Mass the beaker + dry copper. Discard the copper in the place designated by your teacher. Wash your beaker and let dry.

### Data:

|                                           |  |
|-------------------------------------------|--|
| Mass 250 mL beaker                        |  |
| Mass 250 mL beaker + copper (II) chloride |  |
| Mass nails before reaction                |  |
| Mass nails after reaction                 |  |

# Modeling Chemistry Nail Lab

**LL Leslie**



## **Modeling Chemistry Nail Lab:**

## **Modeling Chemistry Nail Lab** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Modeling Chemistry Nail Lab**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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