

## EASEL BALANCING & REACTION QUIZ

Multiple Choice: Select the most correct answer. (15 R)

1. State the type of reaction:  $3 \text{AgNO}_3 + \text{K}_3\text{PO}_4 \rightarrow \text{Ag}_3\text{PO}_4 + 3 \text{KNO}_3$

- A) synthesis                      B) decomposition                      C) single displacement  
D) double displacement                      E) combustion                      F) no reaction

2. State the type of reaction:  $\text{Ca} + 2 \text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2$

- A) synthesis                      B) decomposition                      C) single displacement  
D) double displacement                      E) combustion                      F) no reaction

3. State the type of reaction:  $2 \text{NaClO}_3 \rightarrow 2 \text{NaCl} + 3 \text{O}_2$

- A) synthesis                      B) decomposition                      C) single displacement  
D) double displacement                      E) combustion                      F) no reaction

4. State the type of reaction:  $\text{P}_4 + 5 \text{O}_2 \rightarrow 2 \text{P}_2\text{O}_5$

- A) synthesis                      B) decomposition                      C) single displacement  
D) double displacement                      E) combustion                      F) no reaction

5. Which of the following type(s) of chemical reactions produces water and an ionic compound?

- A) single displacement                      B) neutralization                      C) decomposition                      D) synthesis

6. Chemical equations must be balanced to satisfy \_\_\_\_\_.

- A) the law of definite proportions                      C) the law of conservation of mass  
B) the law of multiple proportions                      D) Avogadro's principle

7. To balance a chemical equation, it may be necessary to adjust the:

- A) coefficient(s)                      B) subscript(s)                      C) formulae of the product(s)                      D) number of products

8. Propane gas,  $\text{C}_3\text{H}_8$ , burns in excess oxygen gas. When the equation for this reaction is correctly balanced and all coefficients are reduced to their lowest whole-number terms, the coefficient for  $\text{CO}_2$  is:

- A) 1                      B) 2                      C) 3                      D) 4                      E) 5                      F) 6

9. Which equation is not balanced?

- A)  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$                       B)  $4\text{H}_2 + 3\text{O}_2 \rightarrow 4\text{H}_2\text{O}$   
C)  $\text{H}_2 + \text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{H}_2\text{O}$                       D)  $2\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$

10.  $\text{C}_6\text{H}_8\text{O}_4\text{S}_2 + \text{O}_{20} \rightarrow \text{CO}_2 + \text{SO}_2 + \text{H}_2\text{O}$

When the equation above is balanced and all coefficients are reduced to their lowest whole-number terms, the coefficient for  $\text{C}_6\text{H}_8\text{O}_4\text{S}_2$  is:

- A) 1                      B) 2                      C) 6                      D) 10                      E) 12

# Mixed Type Practice For Predicting Reactions Answers

**Albert A Gayle**



**Mixed Type Practice For Predicting Reactions Answers:**

The book delves into Mixed Type Practice For Predicting Reactions Answers. Mixed Type Practice For Predicting Reactions Answers is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Mixed Type Practice For Predicting Reactions Answers, encompassing both the fundamentals and more intricate discussions.

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    - Chapter 1: Introduction to Mixed Type Practice For Predicting Reactions Answers
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    - Chapter 3: Mixed Type Practice For Predicting Reactions Answers in Everyday Life
    - Chapter 4: Mixed Type Practice For Predicting Reactions Answers in Specific Contexts
    - Chapter 5: Conclusion
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  3. In chapter 2, this book will delve into the foundational concepts of Mixed Type Practice For Predicting Reactions Answers. This chapter will elucidate the essential principles that must be understood to grasp Mixed Type Practice For Predicting Reactions Answers in its entirety.
  4. In chapter 3, this book will examine the practical applications of Mixed Type Practice For Predicting Reactions Answers in daily life. The third chapter will showcase real-world examples of how Mixed Type Practice For Predicting Reactions Answers can be effectively utilized in everyday scenarios.
  5. In chapter 4, the author will scrutinize the relevance of Mixed Type Practice For Predicting Reactions Answers in specific contexts. The fourth chapter will explore how Mixed Type Practice For Predicting Reactions Answers is applied in specialized fields, such as education, business, and technology.
  6. In chapter 5, this book will draw a conclusion about Mixed Type Practice For Predicting Reactions Answers. This chapter will summarize the key points that have been discussed throughout the book.
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