

**ঘের ও কীটপতঙ্গের অংশের নির্ধারিত পদ্ধতির প্রয়োগের**

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{16.5} = 1.21$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

চলকের ভর,  $m = 20 \text{ g}$

HCl এর আপেক্ষিক ভর,

$$M = 36.5 \text{ g/mol}$$

চলকের আয়তন,  $V = 1 \text{ L}$

চলকের ঘনমাত্রা  $S = ?$

উদাহরণে দ্রবীভূত X-পদার্থিতে 10 mg এর মধ্যে কত পরিমাণ দ্রব ও Y-পদার্থিতে 0.5 g 2%  $\left(\frac{W}{V}\right)$  HCl দ্রবের সাথে মিশ্রিত করা হয়েছে।

যদিও X-পদার্থটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও Y-পদার্থটি হলো ছোটো আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব। নিম্নে তা বিশ্লেষণ করা হলো-

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও পরিমাপক বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে দূষণ উল্লেখযোগ্যভাবে কমে আসে।

এছাড়া সেমি-মাইক্রো পদ্ধতিতে  $\text{H}_2\text{S}$  এর পরিবর্তে ক্রোমোস্টাটাইজ ব্যবহার করা হয় ফলে পরিবেশে দূষণ ঘটে না।

$(\text{CN})_2 + \text{H}_2\text{O} \rightarrow \text{HCN} + \text{HCN} + \text{H}_2\text{S}$

অতএব, কোনো কোনো ক্ষেত্রে কম কষ্টের রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে দূষণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

**উদাহরণ**



পাত্র-ক



পাত্র-খ



পাত্র-ঘ

কি. কে. ১০। এর ক. ১/

১. উদাহরণে দ্রব কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব কেন?
৩. চলকের জন্য প্রয়োজ্য হ্যাঙ্কার প্রতীকটি অঙ্কন করো এবং ব্যাখ্যা করো।
৪. উদাহরণের অংশের দ্রব থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে 1M ঘনমাত্রার দ্রব তৈরি করার প্রক্রিয়া ব্যাখ্যা করো। ৪

৪ নং প্রশ্নের উত্তর

১. বিশ্লেষণের নিমিত্ত বীমের উপর রাইজার স্থাপন করলে বীমের প্রতি স্কেলের জন্য যে ভর পাওয়া যায়, তাকে রাইজার দ্রবক বলে।

২. সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও পরিমাপক বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে দূষণ উল্লেখযোগ্যভাবে কমে আসে। অতএব, কোনো কোনো ক্ষেত্রে কম কষ্টের রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে দূষণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

৩. উদাহরণের দ্রবপট্ট হলো সালফিউরিক এসিড ( $\text{H}_2\text{SO}_4$ ) এর 10 M দ্রব।  $\text{H}_2\text{SO}_4$  একটি জারক ও ক্ষয়কারী পদার্থ।  $\text{H}_2\text{SO}_4$  এর 10 M দ্রব যখন পাত্র  $\text{H}_2\text{SO}_4$  দ্রব। পাত্র  $\text{H}_2\text{SO}_4$  পদার্থকে ক্ষয়কারী হ্যাঙ্কার প্রতীক অঙ্কন করা হয়। পাত্র  $\text{H}_2\text{SO}_4$  এর জন্য ক্ষয়কারী হ্যাঙ্কার প্রতীক নিয়ে অঙ্কন করে ব্যাখ্যা করা হলো:

প্রতীক	ক্ষয়কারী (Corrosive)
আবশ্য	ক্ষয়কারী রাসায়নিক উপাদান কলতে যে ক্ষয়জনক রাসায়নিক দ্রব্যকে খুবই দ্রুত ক্ষয় সম্পর্কে ত্বক ও অন্যান্য পদার্থের ক্ষয়জনক ক্ষতি সঞ্চিত হয়। এতে ত্বক পুড়ে যায় এবং অন্যান্য পদার্থ গলে যায়।
উদাহরণ	পাত্র কঠিন এসিডসমূহ ( $\text{HCl}$ , $\text{HNO}_3$ , $\text{H}_2\text{SO}_4$ ), ক্ষয়কারী ( $\text{NaOH}$ , $\text{KOH}$ ), $\text{H}_2\text{O}_2$ , অক্সিজেন লিকার প্রভৃতি দ্রব ইত্যাদি।
সংরক্ষণ পদ্ধতি	<ul style="list-style-type: none"> <li>পাত্র এসিডকে পোতলে রাখতে হবে।</li> <li>পাত্র সংরক্ষণে, এরপর পাত্রের পিছন থেকে পোতলে রাখতে হবে।</li> </ul>

উদাহরণের দ্রবপট্ট হলো 10M  $\text{H}_2\text{SO}_4$  দ্রব এবং এর থেকে 1M ঘনমাত্রার দ্রব প্রস্তুত করতে হবে 1ম ও ২ম চলকের ব্যবহার করতে হবে।

1ম চলকের দ্রব হলো 100 ml অক্সিজেনমুক্ত জল ও এর 10 ml পিগেট। যেহেতু অক্সিজেনমুক্ত জলের অংশ 100 ml, তাই আমাদের 100 ml 1M ঘনমাত্রার  $\text{H}_2\text{SO}_4$  দ্রব তৈরি করতে হবে। দ্রবের অনুকরণ দ্রব হতে আমরা জানি,

$$V_1 S_1 = V_2 S_2$$

$$\text{বা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left( \frac{100 \times 1}{10} \right) \text{ ml}$$

$$= 10 \text{ ml}$$

অতএব, 10ml  $\text{H}_2\text{SO}_4$  দ্রবের সঙ্গে পানি যোগ করতে হবে

$$(100 - 10) \text{ ml} = 90 \text{ ml}$$

অতএব, পিগেটের সাহায্যে 10 ml দ্রব ঘনমাত্রার দ্রব নিয়ে তার সাথে অক্সিজেনমুক্ত জল 90 ml পানি যোগ করলে আমাদের কামিষ্ঠ ঘনমাত্রার দ্রব পাওয়া যাবে।

**উদাহরণ**



চিত্র-A



চিত্র-B



চিত্র-C

কি. কে. ১০। এর ক. ১/

১. অক্সিজেনমুক্ত জল কত?
২. সামান্য দ্রব K<sub>2</sub> এর ঘন কখনো অসীম হয় না-ব্যাখ্যা করো।
৩. A চিত্রের হ্যাঙ্কারে উত্তর করার কৌশল বর্ণনা করো।
৪. B ও C চিত্রের সাথে সম্পর্কিত রাসায়নিক দ্রব্যের ব্যবহারকালে খান্দা কৃতি দ্রব কবীর উপর সম্পর্কে বিশ্লেষণ করো।

৪ নং প্রশ্নের উত্তর

১. অক্সিজেনমুক্ত জল এক ধরনের রাসায়নিক বা কৃত্রিম দ্রব্য সঞ্চেপক বা দ্রব বা উপাদানসমূহকে জারণ প্রক্রিয়ায় জাল থেকে ত্বক করে থাকার সঞ্চেপক করে।

২. সামান্য দ্রব K<sub>2</sub> এর ঘন কখনো অসীম বা শূন্য হতে পারে না। উভয়টি ব্যাখ্যা করার জন্য নিম্নের উত্তরদ্রব্য মাইক্রোপট বিশ্লেষণ করি-  

$$A + B \rightarrow C + D$$

# Hsc Chemistry 1 First Chapter Notes

**Jim Stamell**



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