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Chapter 3,

Finding Percent by Mass

Carbon	8.44 g carbon	$\frac{8.44 \text{ g carbon}}{20.00 \text{ g sucrose}} \times 100\% =$   42.2% carbon
Hydrogen	1.30 g hydrogen	$\frac{1.30 \text{ g hydrogen}}{20.00 \text{ g sucrose}} \times 100\% =$   6.50% hydrogen
Oxygen	10.26 g oxygen	$\frac{10.26 \text{ g oxygen}}{20.00 \text{ g sucrose}} \times 100\% =$   51.30% oxygen
Total	20.00 g sucrose	 42.2% 51.30% 6.50%

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