

GRAPHING PERIODIC TRENDS

Standard: Students know how to use the periodic table to identify trends in ionization energy, electronegativity, and the relative sizes of ions and atoms.

PRE-LAB DISCUSSION:

The Periodic Table is arranged according to the Periodic Law. The Periodic Law states that when elements are arranged in order of increasing atomic number, their physical and chemical properties show a periodic pattern. Students can discover these patterns by examining the changes in properties of elements on the Periodic Table. The properties that will be examined in this lesson are: atomic radius AND first ionization energy.

PURPOSE:

To understand periodic trends relating to atomic radius, ionization energy, and electronegativity.

PROCEDURE:

Use the information in these tables to complete the graph as described below.

RESULTS:

Symbol	Atomic Radius (Picometers)	First Ionization (Kilojoules/mole)	Electronegativity (4-point scale)
H	37	1312	2.1
Li	128	520	1.0
Na	190	496	0.9
K	203	419	0.8
Rb	220	403	0.8
Cs	264	376	0.7

Symbol	Atomic Radius (Picometers)	First Ionization (Kilojoules/mole)	Electronegativity (4-point scale)
He	31	498	0.0
Mg	141	738	1.2
Al	121	578	1.5
Si	111	787	1.8
P	107	1012	2.1
S	106	1000	2.5
Cl	102	1251	3.0
Ar	100	1521	—

1. Define each of these terms: Atomic radius, First Ionization Energy, and Electronegativity.
2. Using colored-pencil or pen, list under the symbol (in this order) the Atomic Radius, First Ionization Energy, and Electronegativity. Use a different color for each property. Example: Write all of the Atomic radius values in red, all of the First Ionization energies in green, and all of the Electronegativities in blue. Of course, you can pick any colors that are available, as long as you are consistent.
3. Observe the trends in each property as you go down the Alkali metal group, and as you go across Period 3.
4. Write out each of the statements written on the opposite side, completing each statement with the observed trend (increase or decrease).
5. There is NO ABSTRACT to write for this activity :)

Graphing Periodic Relationships Lab Answers

Alex A.T. Bui, Ricky K. Taira



Graphing Periodic Relationships Lab Answers:

Essentials of Experimental Chemistry for Laboratory and Demonstration Alfred Benjamin Garrett, 1959 Laboratory Manual Jo A. Beran, 1990 **General, Organic, and Biological Chemistry Study Guide and Selected Solutions** Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems

Mathematical Foundations and Applications of Graph Entropy Matthias Dehmer, Frank Emmert-Streib, Zengqiang Chen, Xueliang Li, Yongtang Shi, 2017-09-12 This latest addition to the successful Network Biology series presents current methods for determining the entropy of networks making it the first to cover the recently established Quantitative Graph Theory An excellent international team of editors and contributors provides an up to date outlook for the field covering a broad range of graph entropy related concepts and methods The topics range from analyzing mathematical properties of methods right up to applying them in real life areas Filling a gap in the contemporary literature this is an invaluable reference for a number of disciplines including mathematicians computer scientists computational biologists and structural chemists

Exploring Physical Science in the Laboratory John T. Salinas, 2019-02-01 This full color manual is designed to satisfy the content needs of either a one or two semester introduction to physical science course populated by nonmajors It provides students with the opportunity to explore and make sense of the world around them to develop their skills and knowledge and to learn to think like scientists The material is written in an accessible way providing clearly written procedures a wide variety of exercises from which instructors can choose and real world examples that keep the content engaging Exploring Physical Science in the Laboratory guides students through the mysteries of the observable world and helps them develop a clear understanding of challenging concepts

Analytical Chemistry Bryan M. Ham, Aihui MaHam, 2024-02-28 ANALYTICAL CHEMISTRY Detailed reference covering all aspects of working in laboratories including safety fundamentals of analytical techniques lab instrumentation and more A comprehensive study of analytical chemistry as it pertains to the laboratory analyst and chemist Analytical Chemistry begins with an introduction to the laboratory environment including safety glassware common apparatuses and lab basics and continues on to guide readers through the fundamentals of analytical techniques such as spectroscopy and chromatography and introduce examples of laboratory programs such as Laboratory Information Management Systems LIMS This newly updated and revised Second Edition of Analytical Chemistry offers expanded chapters with new figures and the latest developments in the field Included alongside this new edition is an updated companion teaching reference and toolkit program called ChemTech Conveniently available via either app or browser the ChemTech program contains exercises that highlight and review topics covered in the book and features useful calculators and programs including solution makers graphing tools and more To aid in reader comprehension the program also includes an interactive periodic table and chapter summaries Written by two highly qualified authors with

significant experience in both practice and academia Analytical Chemistry covers sample topics such as Basic mathematics in the laboratory including different units the metric system significant figures scientific calculators and ChemTech conversion tools Analytical data treatment including errors in the laboratory precision versus accuracy normal distribution curves and determining errors in methodology Plotting and graphing including graph construction curve fitting graphs of specific equations least squares method and computer generated curves Ultraviolet visible UV Vis spectroscopy including wave and particle theory of light light absorption transitions the color wheel and pigments With complete coverage of the practical aspects of analytical chemistry Analytical Chemistry prepares students for a rewarding career as a chemist or a laboratory technician Thanks to ChemTech integration the book is also a useful and accessible reference for the established chemist or technician already working in the laboratory

Chemistry Alfred Benjamin Garrett,1957 **Technical Reports**

Awareness Circular : TRAC. ,1987-08 *Chemistry* Frank Jenkins,1992 *Automata, Languages and Programming* Pierpaolo Degano,1997-06-18 This book constitutes the refereed proceedings of the 24th International Colloquium on Automata Languages and Programming ICALP 97 held in Bologna Italy in July 1997 ICALP 97 celebrated the 25th anniversary of the European Association for Theoretical Computer Science EATCS which has sponsored the ICALP meetings since 1972 The volume presents 73 revised full papers selected from a total of 197 submissions Also included are six invited contributions ICALP is one of the few flagship conferences in the area The book addresses all current topics in theoretical computer science *Current Hydraulic Laboratory Research in the United States* ,1958 Calculus for the Life Sciences Frederick R. Adler,Miroslav Lovric, *Physical Science* ,2015-03-16 Physical Science for grades 5 to 12 is designed to aid in the review and practice of physical science topics Physical Science covers topics such as scientific measurement force and energy matter atoms and elements magnetism and electricity The book includes realistic diagrams and engaging activities to support practice in all areas of physical science The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series is aligned to current science standards **The Zinc and Iodine Book** Stephen DeMeo,2019-05-15 This book is for chemistry teachers who are thinking about reinventing their laboratory experiments that they provide to their students More than a collection of experiments it is an example of using a chemical theme to teach chemistry Instead of introducing many different chemicals per experiment as is the norm in most lab manuals this novel resource focuses on two commonly found elements Zinc and Iodine So what is so special about these elements At the heart of this resource is a colorful cyclic reaction between zinc and iodine one that produces a compound that can decompose back to its original elements This unique phenomenon demonstrates that matter not only changes but is also conserved through a chemical reaction Knowing that a compound can

be the same but different than the reactants that formed it is to understand the essence of chemical change Complementing this reaction this book contains experimental activities that utilize the zinc and iodine theme to scaffold new concepts such as the properties of matter solid and gas stoichiometry equilibrium kinetics acid base chemistry and electrochemistry This teacher tested resource focuses on a set of safe substances that are appropriate for high school teachers who provide an advanced chemistry placement course and for college instructors teaching a first year chemistry laboratory sequence

Medical Imaging Informatics Alex A.T. Bui,Ricky K. Taira,2009-12-01 Medical Imaging Informatics provides an overview of this growing discipline which stems from an intersection of biomedical informatics medical imaging computer science and medicine Supporting two complementary views this volume explores the fundamental technologies and algorithms that comprise this field as well as the application of medical imaging informatics to subsequently improve healthcare research Clearly written in a four part structure this introduction follows natural healthcare processes illustrating the roles of data collection and standardization context extraction and modeling and medical decision making tools and applications Medical Imaging Informatics identifies core concepts within the field explores research challenges that drive development and includes current state of the art methods and strategies **Advanced Instrumental Methods of Chemical Analysis**

Jaroslav Churáček,1993 **New Frontiers in Stable Isotopic Research** Geological Survey (U.S.),1989 Results of a workshop held in Reston Va January 12 13 1988 on laser microprobe and ion microprobe analysis techniques performed on geological samples for stable and rare gas isotopic ratios **U.S. Geological Survey Bulletin** Samuel Warren Hobbs,William Henry Hays,1983 *U.S. Geological Survey Bulletin* ,1983 **Resources in Education** ,1988 Serves as an index to Eric reports microform

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