

Name _____

Molecular Geometries Lab Report Sheet

Formula	Lewis Structure	No of electron domains/lig	Electron domain geometry	VSEPR Molecular geometry	Hybridization of central atom	Polar ?
CO_2	$\text{O}=\text{C}=\text{O}$ 24 valence 24 2 charge clouds	0	Linear 180°	Linear 	sp	NP
BF_3	$\text{F}-\text{B}-\text{F}$ 24 valence 32 3 charge clouds	0	Trigonal planar 120°		sp ²	NP
SnCl_2	$\text{Cl}-\text{Sn}-\text{Cl}$ 16 valence 24 2 charge clouds	3	Bent <120°		sp ²	P
BF_4^-	$\left[\text{F}-\text{B}-\text{F} \right]^-$ 24 valence 40 4 charge clouds	0	Tetrahedral 109.5°		sp ³	P
AsH_3	$\text{H}-\text{As}-\text{H}$ 8 valence 26 3 charge clouds	4	Trigonal pyramidal 109.5°		sp ³	P
H_2S	$\text{H}-\text{S}-\text{H}$ 8 valence 26 2 charge clouds	2	Bent 109.5°		sp ³	NP
NH_2^-	$\left[\text{H}-\text{N}-\text{H} \right]^-$ 8 valence 10 2 charge clouds	2	Bent 109.5°		sp ³	P
POF_3	$\text{F}-\text{P}-\text{Cl}$ 10 valence 32 4 charge clouds	0	Trigonal bipyramidal 120° + 90°		sp ³ d	P

Geometric Structure Of Molecules Lab Answers

Jessica J Manson



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Laboratory Manual for Principles of General Chemistry Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures

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