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**M Woodhall** 

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web draw the lewis structure for each of the following compounds using vsepr determine the electron pair geometry the molecular geometry and the bond angle in degrees for each compound

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web use the vsepr to determine the electronic and molecular geometries of the structure an example of drawing a lewis dot structure for nf 3 is as follows 1 determine the total number of valence electrons that must appear in the structure **experiment 10 lewis dot structures and molecular geometry** - Aug 03 2022

web this effect determines the atom s geometry and bond angles two regions will be 180 apart three regions will be 120 apart and four regions will be 109 5 apart geometry determination a determine the lewis dot structure of the molecule or ion b for each central atom in the structure determine the areas of electron density that lie

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web degree of reaction zero reaction stage let us first discuss the special case of zero reaction according to the definition of reaction when  $\Lambda$  0 equation upper reveals that h1 h2 and equation lower that  $\beta$ 1  $\beta$ 2 fifty percent reaction stage  $\beta$ 2 and the velocity diagram is symmetrical because of symmetry it is also

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web the 50 reaction turbine has been used widely and has special significance the velocity diagram of a 50 reaction is symmetrical and for the maximum utilization factor the exit velocity v2 must be axial figure 2 40 shows a velocity diagram of a 50 reaction turbine and the effect on the utilization factor

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web p1v1 u1 v12 2 gz1 q p2v2 u2 v22 2 gz2 w p1v1 u1 h1 enthalpy of steam at inlet p2v2 u2 h2 enthalpy of steam at exit h1 v12 2 gz1 q h2 v22 2 gz2 w generally changes in potential energy are negligible z1 z2 if no heat is supplied from the surroundings then q 0

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