

Graphing Periodic Trends Lab

Name _____ Hr _____

PURPOSE: To determine how certain properties are periodic when the elements are arranged in periods or groups.

DISCUSSION: There are many periodic trends that occur as you cross a period or go down a group/family on the Periodic Table. Graphing various elemental properties will help to illustrate these trends.

REFERENCE:

Angstrom - A unit of distance equal to one ten-billionth of a meter.

Atomic Radius - The distance from the center of the nucleus to the valence electron energy level.

Ionization energy - The energy required to remove an electron from an atom.

DATA

Properties of the First 18 Elements

Element	Atomic Number	Atomic Radius (Angstroms)	Ionization Energy
H	1	0.79	314
He	2	0.49	567
Li	3	2.05	124
Be	4	1.4	215
B	5	1.17	191
C	6	0.91	260
N	7	0.75	335
O	8	0.65	314
F	9	0.57	402
Ne	10	?	?
Na	11	2.23	119
Mg	12	1.72	176
Al	13	1.82	138
Si	14	1.46	168
P	15	1.23	242
S	16	1.09	239
Cl	17	0.97	299
Ar	18	?	?

Using the data above, make a line graph the following data. Draw a dark vertical line to separate the 3 periods. Give each graph a title and label both the x-axis and the y-axis. Connect your points. Predict the plot points for elements with question marks.

1. Atomic Radius (y) vs. Atomic Number (x) for the first 3 periods (18 elements – 3 graphs)
2. Ionization energy (y) vs. Atomic Number (x) for the first 3 periods (18 elements – 3 graphs)

Graphing Trends In The Periodic Table Lab Answers

Karen Timberlake



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The Science Teachers Bulletin ,1989 General, Organic, and Biological Chemistry Study Guide and Selected Solutions Karen C. Timberlake,2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems **Prentice Hall Science Explorer** Michael J. Padilla,2002 *Laboratory Manual* Jo A. Beran,1990

Prentice Hall Chemistry ,2000 **Chemistry** Karen Timberlake,1999 Suitable for one or two term lab courses covering general organic and biological chemistry this new edition written by Karen Timberlake features many improvements to the insightful experiments that have made it the leading lab manual Each experiment encourages critical thinking with laboratory goals discussion of related concepts clear instructions new pre lab questions and comprehensive report pages Forty one experiments illustrate the basic principles of chemistry Higher Chemistry: Practice Papers for SQA Exams Barry McBride,2017-10-30 Practise for your SQA exams with three specially commissioned Hodder Gibson Practice Exam Papers Practise with model papers written and checked by experienced markers and examiners Get extra advice with specially written study skills guidance sections Gain vital extra marks and avoid common mistakes with examiner tips

Computer Based Projects for a Chemistry Curriculum Thomas J. Manning,Aurora P. Gramatges,2013-04-04 This e book is a collection of exercises designed for students studying chemistry courses at a high school or undergraduate level The e book contains 24 chapters each containing various activities employing applications such as MS excel spreadsheets and Spartan computational modeling Each project is explained in a simple easy to understand manner The content within this book is suitable as a guide for both teachers and students and each chapter is supplemented with practice guidelines and exercises Computer Based Projects for a Chemistry Curriculum therefore serves to bring computer based learning a much needed addition in line with modern educational trends to the chemistry classroom **Exploring Physical Science in the**

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and skills into everyday life **A World From Dust** Ben McFarland, 2016-03-07 A World From Dust describes how a set of chemical rules combined with the principles of evolution in order to create an environment in which life as we know it could unfold Beginning with simple mathematics these predictable rules led to the advent of the planet itself as well as cells organs and organelles ecosystems and increasingly complex life forms McFarland provides an accessible discussion of a geological history as well describing how the inorganic matter on Earth underwent chemical reactions with air and water allowing for life to emerge from the world's first rocks He traces the history of life all the way to modern neuroscience and shows how the bioelectric signals that make up the human brain were formed Most popular science books on the topic present either the physics of how the universe formed or the biology of how complex life came about this book's approach would be novel in that it condenses in an engaging way the chemistry that links the two fields This book is an accessible and multidisciplinary look at how life on our planet came to be and how it continues to develop and change even today This book includes 40 illustrations by Gala Bent print artist and studio faculty member at Cornish College of the Arts and Mary Anderson medical illustrator The Software Encyclopedia 2001 ,2001 Chemistry McGraw-Hill Staff, 2001-07 **Nuclear Science Abstracts** ,1961 Holt Science & Technology Tennessee Holt Rinehart & Winston, 2003 *Chemistry* Martin Stuart Silberberg, 1996 This general chemistry text centres on the theme that observable change in chemical systems is the result of molecular change The aims of this edition are to enable students to perceive matter and change at the molecular level and to help build student confidence in their ability to solve chemical problems as they discover the relevance of chemistry to their lives *ERDA Energy Research Abstracts* United States. Energy Research and Development Administration, 1976

Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 *Physical Properties of Materials, Third Edition* Mary Anne White, 2018-10-12 Designed for advanced undergraduate students and as a useful reference book for materials researchers *Physical Properties of Materials Third Edition* establishes the principles that control the optical thermal electronic magnetic and mechanical properties of materials Using an atomic and molecular approach this introduction to materials science offers readers a wide ranging survey of the field and a basis to understand future materials The author incorporates comments on applications of materials science extensive references to the contemporary and classic literature and 350 end of chapter problems In addition unique tutorials allow students to apply the principles to understand applications such as photocopying magnetic devices fiber optics and more This fully revised and updated Third Edition includes new materials and processes such as topological insulators 3 D printing and more information on nanomaterials The new edition also now adds Learning Goals at the end of each chapter and a Glossary with more than 500 entries for quick reference **Physical Activity Assessments for Health-related Research** Greg Welk, 2002 And examples References Construct validity in physical activity research Matthew T Mahar and David A Rowe Definitional stage Confirmatory stage Theory testing stage Summary References Physical activity data odd distributions yield strange answers Jerry R Thomas and

Katherine T Thomas Overview of the general linear model and rank order procedures Determining whether data are normally distributed Application of rank order procedures Data distributions and correlation Extensions of GLM rank order statistical procedures Summary Endnote References Equating and linking of physical activity questionnaires Weimo Zhu What is scale equating Equating methods Practical issues of scale equating Remaining challenges and future research directions Summary References

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