

TITRATIONS



Calculate the concentration of the sulfuric acid in mol/dm³

Use only the student's concordant results

mean titre = $\frac{22.10 + 22.15 + 22.15}{3} = 22.13 \text{ cm}^3$ ✓

$n(\text{NaOH}) = 0.105 \times (22.13/1000) = 0.002324 \text{ mol}$ ✓

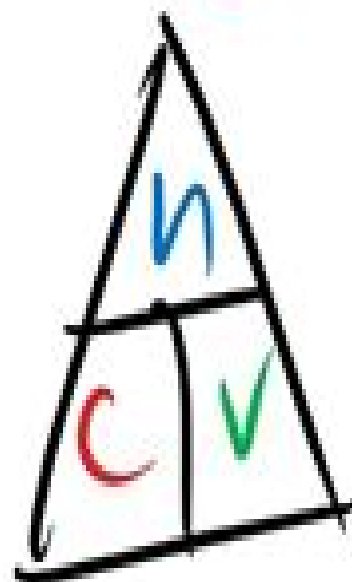
$n(\text{H}_2\text{SO}_4) = 0.002324 / 2 = 0.001162 \text{ mol}$ ✓

$c(\text{H}_2\text{SO}_4) = \frac{0.001162}{(25/1000)} = 0.0465 \text{ mol/dm}^3$ ✓

$$n = m/M_r$$

$$n = c \times v$$

$$c = n/v$$



GCSE CHEMISTRY

Gcse Titration Practise Questions

A Gutmann



Gcse Titration Practise Questions:

Gcse Titration Practise Questions Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has be much more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Gcse Titration Practise Questions**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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