- 2 -

- Which of the following processes provide evidence of the particulate nature of matter?
 - I. Capillarity
 - II. Diffusion
 - III. Filtration
 - IV. Osmosis
 - (A) I and III only
 - (B) I and IV only
 - (C) II and III only
 - (D) II and IV only
- 'Mass number' is the number of
 - (A) neutrons plus protons
 - (B) neutrons minus protons
 - (C) electrons plus neutrons
 - (D) electrons plus protons
- 3. Which of the following elements has 7 electrons in its outer shell?
 - (A) Hydrogen
 - (B) Oxygen
 - (C) Nitrogen
 - (D) Chlorine
- An atom that has 20 electrons when it ionizes has the same electronic configuration as
 - (A) argon
 - (B) calcium
 - (C) neon
 - (D) sodium
- ³⁴Na and ²⁵Na are isotopes because they
 - (A) conduct electricity
 - (B) react vigorously with water
 - C) have the same number of protons
 - (D) have the same relative atomic mass

- Sulfur and oxygen are in the same group of the periodic table because
 - (A) they can react with each other
 - (B) they can form covalent compounds
 - (C) they have the same number of electrons in their outer shell
 - (D) the atomic number of sulfur is 16 and the relative atomic mass of oxygen is 16

<u>Item 7</u> refers to the following quantities of atoms.

- 12 g of carbon
- II. 16 g of sulfur
- III. 23 g of sodium
- IV. 12 g of magnesium

(Relative mass number: C = 12; S = 32; Mg = 24; Na = 23)

- 7. Which of the following pairs represents equal quantities of atoms?
 - (A) I and II only
 - (B) I and IV only
 - (C) II and III only
 - (D) II and IV only
- An ion with a single negative charge may be converted into a neutral atom by
 - (A) losing a neutron
 - (B) gaining a proton
 - (C) losing an electron
 - (D) gaining an electron

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