

Gases

Q2a

Polytropic expansion according to $PV^{1.2}$ is encountered by a gas in piston-cylinder from a pressure of 1.5 MPa and an initial volume of 0.25 m³ to a volume of 0.3 m³. Take the characteristic gas constant as 0.288 kJ/kg.K and the specific heat capacity at constant volume as 0.717 kJ/kg.K.

Calculate each of the following:

- | | | |
|-------|--|------|
| 2.1 | Pressure after expansion | (3) |
| 2.2 | Work done | (4) |
| 2.3 | Ratio of $\frac{T_2}{T_1}$ | (2) |
| 2.4 | Assume a mass of 1 kg.
Determine each of the following: | |
| 2.4.1 | T_1 and T_2 | (5) |
| 2.4.2 | Internal energy | (3) |
| 2.4.3 | Heat energy | (3) |
| | | [20] |

Q2b

A balloon is filled with 0.5 kg gas at constant pressure from a temperature of 24 °C and a volume of 0.05 m³ to a volume of 0.1 m³. Take the characteristic gas constant as 0.3 kJ/kg.K and the specific heat capacity at constant pressure as 1.005 kJ/kg.K

Calculate each of the following:

- | | | |
|-----|--|------|
| 2.1 | Pressure at end of constant pressure process | (3) |
| 2.2 | Final temperature | (3) |
| 2.3 | Work done in kJ | (3) |
| 2.4 | Internal energy in kJ | (4) |
| 2.5 | Heat energy in kJ | (3) |
| 2.6 | Change in entropy | (4) |
| | | [20] |

Gases Section 1 Answers

M Tight



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