



Directions: Please show all work and round each answer the correct number of significant figures. (2)

1. A gas occupies 100.0 mL at a pressure of 760 mm Hg. What is the volume of the gas when its pressure is increased to 880 mm Hg?
2. A gas has a pressure of 0.450 atm and a volume of 32 L. At what pressure would the volume of the gas change to 28 L?
3. A gas occupies 525 mL at a pressure of 85.0 kPa. What would the volume of the gas be at a pressure of 65.0 kPa?
4. A gas occupies 3.5 L at standard pressure. Find the volume of the gas when the pressure is 1140 mm Hg.
5. A gas has a volume of 750 mL at a pressure of 2.15 atm. What will the pressure be if the volume becomes 1.25 L?

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Rachel Sandford



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Table of Contents Gas Laws Homework Answers

1. Understanding the eBook Gas Laws Homework Answers
 - The Rise of Digital Reading Gas Laws Homework Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Gas Laws Homework Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gas Laws Homework Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gas Laws Homework Answers
 - Personalized Recommendations
 - Gas Laws Homework Answers User Reviews and Ratings
 - Gas Laws Homework Answers and Bestseller Lists
5. Accessing Gas Laws Homework Answers Free and Paid eBooks
 - Gas Laws Homework Answers Public Domain eBooks
 - Gas Laws Homework Answers eBook Subscription Services

- Gas Laws Homework Answers Budget-Friendly Options
- 6. Navigating Gas Laws Homework Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Gas Laws Homework Answers Compatibility with Devices
 - Gas Laws Homework Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gas Laws Homework Answers
 - Highlighting and Note-Taking Gas Laws Homework Answers
 - Interactive Elements Gas Laws Homework Answers
- 8. Staying Engaged with Gas Laws Homework Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gas Laws Homework Answers
- 9. Balancing eBooks and Physical Books Gas Laws Homework Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gas Laws Homework Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gas Laws Homework Answers
 - Setting Reading Goals Gas Laws Homework Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gas Laws Homework Answers
 - Fact-Checking eBook Content of Gas Laws Homework Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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