

Gas Laws Worksheet (Chapter 5)

Boyles Law: $P_1V_1=P_2V_2$
(Inverse Relationship)

Charles' Law: $\frac{V_1}{T_1} = \frac{V_2}{T_2}$
(Direct Relationship)

Combined Gas Law: $\frac{P_1V_1}{T_1} = \frac{P_2V_2}{T_2}$

Ideal Gas Law: $PV=nRT$

Density of a gas : $d = \frac{\text{mass}}{\text{vol}} = \frac{P(\text{MW})}{RT}$

P = Pressure in atmospheres (atm)
1 atm = 760 torr = 760 mm Hg

V = Volume in liters (L)

T = Temperature in Kelvin (K)

n = Number of moles

R = Ideal gas constant
R = 0.082057 L(atm)/K(mol)

Using the proper equations solve (complete) each question. Assume all are "ideal" gasses.

1. What is the pressure of a gas if you compressed the gas from its original 500 mL at 3.4 torr to a volume of 302 mL?
2. At what temperature will a gas be at if you allow it to expand from an original 456 mL at 65 °C to 3.4 L?
3. If you have 0.56 moles of an ideal gas at 87 °C and a pressure of 569 torr, what volume will the gas take up?
4. You have a gas at 453 mm Hg with a volume of 700 mL and a temperature of 25 °C, what will the temperature of the gas be, if you change the pressure to 278 mm Hg and a volume of 1200 mL?
5. Analysis of a gaseous chlorofluorocarbon, CCl_xF_y , shows it contains 11.79 % C and 69.57 % Cl. In another experiment you find that 0.107 g of the compound fills a 458 mL flask at 25 °C with a pressure of 21.3 mmHg. What is the molecular formula of the compound?

Gas Law Worksheet 2 Answers

Albert A Gayle

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name Albert A Gayle.

Gas Law Worksheet 2 Answers:

Unveiling the Energy of Verbal Artistry: An Mental Sojourn through **Gas Law Worksheet 2 Answers**

In some sort of inundated with screens and the cacophony of instant interaction, the profound power and emotional resonance of verbal beauty usually diminish in to obscurity, eclipsed by the continuous onslaught of noise and distractions. However, situated within the lyrical pages of **Gas Law Worksheet 2 Answers**, a charming work of literary elegance that pulses with natural thoughts, lies an remarkable journey waiting to be embarked upon. Penned by a virtuoso wordsmith, this mesmerizing opus courses viewers on a mental odyssey, gently exposing the latent potential and profound impact embedded within the intricate web of language. Within the heart-wrenching expanse of the evocative evaluation, we will embark upon an introspective exploration of the book is central themes, dissect their fascinating publishing style, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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