# 6 • The Gas Laws

#### GAS STOICHIOMETRY WORKSHEET

Please answer the following on separate paper using proper units and showing all work. Please note that these problems require a balanced chemical equation.

- Carbon monoxide reacts with oxygen to produce carbon dioxide. If 1.0 L of carbon monoxide reacts with oxygen at STP.
  - a. How many liters of oxygen are required to react?
  - b. How many liters of carbon dioxide are produced?
- Acetylene gas (C<sub>2</sub>H<sub>2</sub>) undergoes combustion to produce carbon dioxide and water vapor.
  - a. How many liters of C<sub>2</sub>H<sub>2</sub> are required to produce 75.0 L of CO2?
  - b. What volume of H<sub>2</sub>O is produced?
  - c. What volume of O<sub>2</sub> is required?
- 3. If liquid carbon disulfide (CS<sub>2</sub>) reacts with 450 mL of oxygen to produce the gases carbon dioxide and sulfur dioxide, what volume of each product is produced?
- Assume that 5.60 L of hydrogen gas at STP reacts with copper (II) oxide according to the following balanced equation:
  CuO (s) + H₂ (g) → Cu (s) + H₂O (g)
  - a. How many moles of H2 react?
  - b. How many moles of copper are produced?
  - c. How many grams of copper are produced?

# **Gas Stoichiometry Problem And Answer**

**Léo Pomar** 

**Gas Stoichiometry Problem And Answer:** 

Whispering the Techniques of Language: An Mental Quest through Gas Stoichiometry Problem And Answer

In a digitally-driven earth wherever displays reign great and quick conversation drowns out the subtleties of language, the profound techniques and emotional subtleties hidden within phrases frequently go unheard. However, set within the pages of **Gas Stoichiometry Problem And Answer** a interesting literary prize pulsing with organic feelings, lies a fantastic quest waiting to be undertaken. Written by an experienced wordsmith, this wonderful opus invites viewers on an introspective trip, gently unraveling the veiled truths and profound affect resonating within the material of each word. Within the mental depths of the poignant evaluation, we shall embark upon a heartfelt exploration of the book is primary styles, dissect their captivating writing type, and fail to the strong resonance it evokes deep within the recesses of readers hearts.

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