Presence of a Catalyne

- OWhat effect does the MnO₂ catalyst have on the rate of evolution of O₂ gas?
- 2. Write a balanced equation for the decomposition of H₂O₂-

E. Concentration of Reactants: Magnesium-Hydrochloric Acid System

of HCI	of HCI	mol HCI	mass of Mg	O _{mol Mg}	mol HCl mol Mg	O'Time (sec)
6.M	-5m		0.0301	DESCRIPTION OF THE PARTY OF	Section 1	
4 M	5,~	Section 1	0.0188	Constitution of		
3.M	5/2		0.0161	AMERICAN PROPERTY.		
1 M	500		0.0100		or the second	
The second secon						

- 2. From your graph, predict the time, in seconds, for 5 mg of Mg to react in 5 mL of 2.0 M HCL

Concentration of	Reactants: Indi	c Acid-Sulfurous	Archel Continue

Molar concentration of HIO, ______ Molar concentration of H₂SO, ______

What is the volume (mL) per drop of the HIO, solution? _____ Owhat is the volume (mL) per drop of the water? _____

Test Tube	Drops of HIO ₃	mL HIO ₃	Drops of H ₂ O	mL H ₂ O	mL H ₂ SO ₃	[HIO ₃] ₆ ⁶ (diluted)	Timse (sec)
91		-			1.0		
2		THE OWNER OF THE PERSON NAMED IN	PART OF THE PART O	Section 1	1.0	THE RESERVED	HOUSE STREET
3				1	1.0	P. S. P. S. DOWN CO.	
4 .					1.0	THE RESERVE	
5					1.0		

- L. Opiot [HIO₁]₀ (y-axis) versus time (x-axis). Instructor's approval of graph:
- 2. How does a change in the molar concentration of HIO, affect the time required for the appearance of the deep-blue I₁"starch complex?
- 3. Estimate the time, in seconds, for the deep-blue I₁*-starch complex to form when 10 drops of 0.01 M HIO₃ are used for the reaction. Assume all other conditions remain constant.

"See footnote 5

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