Name ______ Period _____ Date _____

Chapter 5: Thermal Energy

Lesson 1: Thermal Energy, Temperature, and Heat

Thermal energy - The sum of kinetic and potential energy of the particles that make up a material

- Thermal energy causes a phase change.
- Thermal energy describes the energy of the particles that make up a solid.
 liquid, or gas.

Thermal Energy - Kinetic

- Every solid, liquid, gas contains trillions of tiny particles that are in motion.
- Because these particles are in motion they all have kinetic energy.
- The <u>faster</u> the particles move, the more <u>kinetic energy</u> they have.
 - o Solids particles vibrate in place
 - Liquids particles move a little more freely
 - Gases particles move most freely and quickly

Thermal Energy - Potential

- Particles that make up <u>solids</u>, <u>liquids</u>, <u>and gases</u> also have <u>potential energy</u> which is the <u>distance between particles</u>.
- The greater the <u>distance</u> between <u>particles</u> the greater the <u>potential energy</u> of the <u>particles</u>.
 - o Solids Particles are held close together by attractive forces
 - Liquids Particles are slightly farther apart
 - o Gases Particles are much more spread out than in those in a solid

Heat - The movement of thermal energy from a warmer object to a cooler object
Why do your hands get warm when you hold a cup of coffee? The thermal energy
from the warm cup is moving to and heating up your cool hands.

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