

+ WS 7.1 Specific Heat & Calorimetry

Show all work neatly.....

$$q = m c \Delta T$$

Specific Heats	
substance	c (J/g°C)
water	4.184
ethanol	2.452
graphite	0.720
diamond	0.502
iron	0.444
copper	0.385
silver	0.237
gold	0.129
ice	2.092

1. How much heat is required to raise the temp of 654 g of water from 34.5°C to 89.7°C?

Ans _____

2. How much heat is required to raise the temp of 654 g of silver from 34.5°C to 89.7°C?

Ans _____

3. If 7350 J were added to 152 g of ethanol, its temp would go up by how much?

Ans _____

4. 16.25 g of water at 54.0°C releases 402.7 J. What will be its final temp?

hint: it's cooling down, so the final temperature will be less than 54.0°C

Ans _____

5. 697 J are added to a 36.8 g of kerosene and the temp increases from 22.5°C to 34.7°C. Determine kerosene's specific heat.

Ans _____

6. 25 copper pennies (each weighing 3.12 g) are placed in 36.0 g of ethanol at room temp (22.1°C). How much heat will it take to raise the temperature up to 65.8°C?

hint: Calculate q for the copper & q for the ethanol separately. Then add your 2 answers together

Ans _____

7. What mass of 54.0°C water must be added to 468 g of 21.0°C water to make the final temp of both come out to be 29.0°C?

Ans _____

8. What mass of 54.0°C gold must be added to 468 g of 21.0°C water to make the final temp of both come out to be 29.0°C?

Ans _____

9. A 325 g brass rod at 100.0°C is placed in a cup containing 162 g of 24.3°C water. The final temp comes out to be 37.4°C. Determine brass's specific heat.

Ans _____

10. 100.0 g of water at 20.0°C are mixed with 200.0 g of copper at 40.0°C. What will the final temp come out to be?

Ans _____

Heat Calculations Practice Answers

Yicheng Fang

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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web oct 22 2020 amazing solo by pat martino over the jazz standard along came betty as usual impressive solo with quick phrases pat martino is an expert in this kind of solos which are very difficult to

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