	lon				
	each sentence or	r statement.			
12.	An electron that of	occupies the outer electron.	most energy l	evel of an ato	m is known as a(n)
13.	An atom or group one or more elect		an electrical	charge becau	se it has either lost or gained
14.	An ion that has a		sa(n) Chr	0-	70 C
	The ion formed by			ation	
16.	The ion formed by	y an atom of a nor	nmetal elemen	it is a(n)	n on
	a(n)5 \(\sim / f \)				veen a cation and an anion is
18.	A repetitive geom	etric arrangement	t of ions, atom	s, or molecule	es that forms a crystal structur
19.	is called the _<	ion O^2 is the	xule.	ion.	
20.	The name of the	ion Cu* is the	0017-(1)	ion.	
21.	The chemical form	mula for the comp	ound strontiun		th contains Sr2+ and S2- ions,
	5-5				
ord Ba					
	Valence	Crystal lattice Salt	Oxide	Anion	Copper(I)
rite the	formula for the f	following tonic	Write th	e names for	
rite the		following lonic	Write th		the following ionic
mpou		Nac1	compou		
mpour 1. So	nds:	Nac1 Cao	compou	ınds:	the following ionic
1. So 2. Ca 3. Iro	nds: dium chloride: dcium oxide: n (II) fluoride:	No.CI CaO Fe F2	сотроі 1 	ınds:	the following ionic
1. So 2. Ca 3. Iro 4. Co	nds: dium chloride: dcium oxide: n (II) fluoride: apper (I) oxide:	Nocl Cao Fe Fz Cuzo	сотроі 1 2	inds: . NaNO ₃	sodium Nitrate
1. So 2. Ga 3. Iro 4. Co 5. Alu	nds: dium chloride: dcium oxide: n (II) fluoride:	No.CI CaO Fe F2	2 (2) (3) (3)	inds: . NaNO3 . Li ₂ O	salium Nitrate
1. So 2. Ca 3. Iro 4. Co 5. Alu	nds: dium chloride: dcium oxide: n (II) fluoride: apper (I) oxide: uminum sulfate:	No.CI CaO Fe F2 Cu20 A12 (50 A5 CI	2 () () () () () () () () () (. NaNO ₃ . Li ₂ O . CuCO ₃	lithium oxile Copper-CII) carbon Iron (III) oxide
1. So 2. Ca 3. Iro 4. Co 5. Alu 6. Sil	nds: dium chloride: dcium oxide: n (II) fluoride: pper (I) oxide: uminum sulfate: ver chloride:	No.Cl Ca O Fe F2 Cu2 O Alz (SO Ag Cl Kz CO; Sn Fy	2 2 2 3 4 5 6	. NaNO ₃ . Li ₂ O . CuCO ₃ . Fe ₂ O ₃	Sodium Nitrate Lithium oxile Copper-CID carbon Iron (NI) oxide
1. So 2. Ca 3. Iro 4. Co 5. Alu 6. Silt 7. Po 8. Tir	nds: dium chloride: dcium oxide: n (II) fluoride: pper (I) oxide: uminum sulfate: ver chloride: dassium carbonate n (IV) fluoride:	NoCl CaO Fe Fz Cuzo Alz (SO Ascl Kz Coz Sn Fy Crzo	compoi 1 2 3 4 5	inds: . NaNO ₃ . Li ₂ O . CuCO ₃ . Fe ₂ O ₃ . Ca ₃ (PO ₄) ₂	the following lonic Salium Nitrate Lithium oxile Copper-CII) carboni Iron (III) oxide Calcium phosphate
1. So 2. Ca 3. Iro 4. Co 5. Alu 6. Silt 7. Po 8. Tir 9. Ch	nds: dium chloride: dicium oxide: n (II) fluoride: apper (I) oxide: uminum sulfate: ver chloride: dassium carbonate n (IV) fluoride:	NoCl CaO Fe Fz Cuzo Al, Cso Ascl KzCo, Sm Fy	compoi 1 2 3 5 6	inds: NaNO ₃ Li ₂ O CuCO ₃ Fe ₂ O ₃ Ca ₃ (PO ₄) ₂ SnO	the following ionic Salium Nitrate Lithium oxile Copper-CII) carbon Iron (III) oxide Calcium phosplate tin (III) oxide
1. So 2. Ca 3. Iro 4. Co 5. Alu 6. Silt 7. Po 8. Tir 9. Ch	nds: dium chloride: dcium oxide: n (II) fluoride: pper (I) oxide: uminum sulfate: ver chloride: dassium carbonate n (IV) fluoride:	NoCl CaO Fe Fz Cuzo Alz (SO Ascl Kz Coz Sn Fy Crzo	compoi 1 2 3 5 6 6	inds: NaNO ₃ Li ₂ O CuCO ₃ Fe ₂ O ₃ Ca ₃ (PO ₄) ₂ SnO MgCO ₃	the following ionic Sodium Nitrate Lithium oxile Copper (11) carbon Iron (111) oxide Calcium phosplate tin (11) oxide magnesium carbonat

Holt Science Spectrum Chemical Reactions Answers

Daniel F McAuley

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