



Chemically combine with H<sub>2</sub>O to create acids.

\* Combuston of South Sale Opposity south

S + O<sub>2</sub> --> SO<sub>2</sub> mark coeffice devotable Constitution 1

 $90_{\gamma} \sim 0_{\gamma} \sim 90_{\gamma}$ 

 $SO_3 + H_2O \longrightarrow H_2SO_4$ conditions acted Combustion of Book Solic Oxymatic goodines

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2NO<sub>2</sub> × H<sub>2</sub>O = HNO<sub>2</sub> + HNO<sub>3</sub>

#### WCTED BOVEN:

- Reduces photosynthesis - Leuches toxic Al ions

Lengther matricines from soil - prevents fish egg hatching (pH 5 or lower in water) - high acid in pond cause fish kills

# Heavy Metals

AS

"Heavy metal poisoning: ingestion through food or relation and installment

\*Austr effects: names, vomiting, abdominal pain, distribute, tingling extremities, shortness of breath, whills, resh

\* Chronic: mood change, irritable, teeth and beir loss, slowed mental capacity, mental retardation IQ drop. and developmental delays in developing children fotos. About the

# CHEMICAL cheat sheet

-1.76 ampoint with small amounts of all other gases including -0.04% greenhouse gases

\*H2O is a variable gas that can make up between 0% -4% of the atmosphere depending on location.

# Atmospheric Gases

21% Attenouphoris Service Services BUILDINGS TO 78%

#### Air Pollutants

#### Outdoor Air Pollutants - Criteria Pollutants

SO<sub>2</sub> - sulfur dioxide

\*main source: coal burning

\*reacts with rain to create acid rain \*part of industrial smog

NO2 - nitrogen dioxide

\*main source: transporta-

tion - gasoline burning \*poisonous gas on its own \* creates acid rain when

reacting with rain water \*part of photochemical smog using UV radiation to produce ozone

On - oxome

highly reactive gas

\*poisonous to breathe \*burns respiratory tract \*generated by photochemicall smog at ground surface.

CO - carbon monoxide \*takes place of oxygen in

blood and deprives body of

oxygen \*high concentrations cause

death

Pb - lead.

\*heavy metal poisoning \*main source: lead smelting and industrial processes

PM - particulate matter solid particles that get blown or burned into the air: soot, sediment, etc.

# Industrial Smog Reactions

Components of industrial smog: carbon dioxide, carbon monoxide, sulfur dioxide, sulfur trioxide, sulfuric acid

earbon dioxide.

 $C = O_0 \longrightarrow CO_0$ 

carbon monoxide

2C + O<sub>2</sub> => 2CO

sulfur compounds

Si - On our SiOn SO<sub>2</sub> moO<sub>2</sub> m SO<sub>3</sub>

SO<sub>3</sub> = H<sub>2</sub>O --- H<sub>2</sub>SO<sub>4</sub>

# Photochemical Smog Reactions

 $N_2 + O_2 -- 2860$ 2500 HOurs 2500y.

NOv = EV radiation --> NO + O

 $O + O_1 \rightarrow O_2$ NO HOUSE NO. HO.

#### Indoor Air Pollutants

Tobacco smoke - #1 cause of lung cancer Radon-222 - #2 cause of lung career. Radioactive isotope from decaying uranium in rocks. Seeps into basements. Asbestos - mineral to used for building materials. Cancerous, causes respiratory illnesses including meso-theliomoa (tumors of the lungs, abdomen, and heart) and

abestosis (scaring of lung tissue). Lead - mainly in food and water, but can be from lead

paint chips and dust.

CO - furnaces, gas stoves, gas driers; deprives body of

VOCs (volatile organic compound) - can be cancerous, in plastics, carpets, acrosol paints, cleaning products, air fresheners, moth balls

Biological threat - mold, dust mites, pet dander, can cause allergies; some molds toxic.

Wood burning - particulate mater, soot, CO.

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**Chandresh Agrawal, Nandini Books** 

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### **Table of Contents Help Sheet Page 101 Science Answer**

- 1. Understanding the eBook Help Sheet Page 101 Science Answer
  - The Rise of Digital Reading Help Sheet Page 101 Science Answer
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Help Sheet Page 101 Science Answer
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Help Sheet Page 101 Science Answer
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Help Sheet Page 101 Science Answer
  - Personalized Recommendations
  - Help Sheet Page 101 Science Answer User Reviews and Ratings

- Help Sheet Page 101 Science Answer and Bestseller Lists
- 5. Accessing Help Sheet Page 101 Science Answer Free and Paid eBooks
  - Help Sheet Page 101 Science Answer Public Domain eBooks
  - Help Sheet Page 101 Science Answer eBook Subscription Services
  - Help Sheet Page 101 Science Answer Budget-Friendly Options
- 6. Navigating Help Sheet Page 101 Science Answer eBook Formats
  - o ePub, PDF, MOBI, and More
  - Help Sheet Page 101 Science Answer Compatibility with Devices
  - Help Sheet Page 101 Science Answer Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Help Sheet Page 101 Science Answer
  - Highlighting and Note-Taking Help Sheet Page 101 Science Answer
  - o Interactive Elements Help Sheet Page 101 Science Answer
- 8. Staying Engaged with Help Sheet Page 101 Science Answer
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Help Sheet Page 101 Science Answer
- 9. Balancing eBooks and Physical Books Help Sheet Page 101 Science Answer
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Help Sheet Page 101 Science Answer
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Help Sheet Page 101 Science Answer
  - Setting Reading Goals Help Sheet Page 101 Science Answer
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Help Sheet Page 101 Science Answer
  - Fact-Checking eBook Content of Help Sheet Page 101 Science Answer
  - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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