

## Specific Heat Capacity - Practice Problems

$$Q = mc\Delta T$$

**Q** = heat energy in Joules (J)

**m** = mass in grams (g)

**c** = specific heat in Joules per gram degree celsicus (J/g·°C)

**ΔT** = "delta T", or temperature **change** in °C or Kelvin

\*Remember:  $\Delta T = (T_{\text{final}} - T_{\text{initial}})$ .

**Show all work and proper units.** (e.g. Given, Find, equation, units, etc... watch sig figs & use scientific notation in your final answer)

1. A 15.75 gram piece of iron absorbs 1086.75 joules of heat energy, and its temperature changes from 25°C to 175°C. Calculate the specific heat of iron.

**Given**

$$m = 15.75 \text{ g}$$

$$q = 1086.75 \text{ J}$$

$$\text{Initial } T = 25^{\circ}\text{C}$$

$$\text{Final } T = 175^{\circ}\text{C}$$

**Find**

$$C = ?$$

**Equation**

$$q = mc\Delta T$$

$$c = q/m\Delta T$$

$$\Delta T = (T_{\text{final}} - T_{\text{initial}})$$

**Solve**

$$\Delta T = (T_{\text{final}} - T_{\text{initial}})$$

$$\Delta T = 175^{\circ}\text{C} - 25^{\circ}\text{C}$$

$$\Delta T = 150^{\circ}\text{C}$$

$$c = 1086.75 \text{ J} / (15.75 \text{ g})(150^{\circ}\text{C})$$

$$= 1086.75 \text{ J} / 2362.5 \text{ g} \times ^{\circ}\text{C}$$

$$= 0.46000 \text{ J/g} \times ^{\circ}\text{C}$$

$$= 4.6000 \times 10^{-1} ^{\circ}\text{C}$$

2. 297 J of heat are needed to raise the temperature of aluminum from 22°C to 55°C where the specific heat of aluminum is 0.90 J/g·°C. What is the mass of the aluminum?

**Given**

$$Q = 297 \text{ J}$$

$$c = 0.90 \text{ J/g} \cdot ^{\circ}\text{C}$$

$$T_{\text{initial}} = 22^{\circ}\text{C}$$

$$T_{\text{final}} = 55^{\circ}\text{C}$$

**Find**

$$m = ?$$

**Equation**

$$m = q / (\Delta T)(c)$$

$$\Delta T = (T_{\text{final}} - T_{\text{initial}})$$

# Heat Change Practice Problems

**Dr. B. Kapila, S. K. Khanna**



## Heat Change Practice Problems:

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**Applied Thermodynamics for Meteorologists** Sam Miller, 2015-06-04 This textbook on atmospheric thermodynamics is for students of meteorology or atmospheric science It also serves as a reference text for working professionals in meteorology and weather forecasting It is unique because it provides complete calculus based derivations of

basic physics from first principles and connects mathematical relationships to real world practical weather forecasting applications Worked examples and practice problems are included throughout

**Solving Direct and Inverse Heat Conduction Problems** Jan Taler,Piotr Duda,2010-04-16 This book is devoted to the concept of simple and inverse heat conduction problems The process of solving direct problems is based on the temperature determination when initial and boundary conditions are known while the solving of inverse problems is based on the search for boundary conditions when temperature properties are known provided that temperature is the function of time at the selected inner points of a body In the first part of the book Chaps 1 5 we have discussed theoretical basis for thermal conduction in solids motionless liquids and liquids that move in time In the second part of the book Chapters 6 26 we have discussed at great length different engineering problems which we have presented together with the proposed solutions in the form of theoretical and mathematical examples It was our intention to acquaint the reader in a step by step fashion with all the mathematical derivations and solutions to some of the more significant transient and steady state heat conduction problems with respect to both the movable and immovable heat sources and the phenomena of melting and freezing Lots of attention was paid to non linear problems The methods for solving heat conduction problems i e the exact and approximate analytical methods and numerical methods such as the finite difference method the finite volume method the finite element method and the boundary element method are discussed in great detail Aside from algorithms applicable computational programs written in a FORTRAN language were given

**Principles of Heat Transfer** Massoud Kaviani,2002 CD ROM contains Equations and relations models for thermal circuit modeling

Basic Principles and Calculations in Chemical Engineering David Mautner Himmelblau,James B. Riggs,2012 Best selling introductory chemical engineering book now updated with far more coverage of biotech nanotech and green engineering Thoroughly covers material balances gases liquids and energy balances Contains new biotech and bioengineering problems throughout

*The Practice of Chemistry Study Guide & Solutions Manual* Pamela Mills,Amina El-Ashmawy,2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

**Chapter-wise NCERT + Exemplar + Practice Questions with Solutions for CBSE Chemistry Class 11 - 2nd Edition** Disha Experts,2017-08-29 The book Chapter wise NCERT Exemplar Practice Questions with Solutions for CBSE Class 11 Chemistry has been divided into 3 parts Part A provides detailed solutions Question by Question of all the questions exercises provided in the NCERT Textbook Part B provides solutions to the questions in the NCERT Exemplar book Part C provides selected Practice Questions useful for the Class 11 examination along with detailed solutions The solutions have been designed in such a manner Step by Step that it would bring 100% Concept Clarity for the student

**General Organic and Biological Chemistry** Kenneth W. Raymond,2009-12-14 This general organic and biochemistry text has been written for students preparing for careers in health related fields such as nursing dental hygiene nutrition medical technology and occupational therapy It is also suited

for students majoring in other fields where it is important to have an understanding of the basics of chemistry Students need have no previous background in chemistry but should possess basic math skills The text features numerous helpful problems and learning features I-physics Iv' 2006 Ed. , Comprehensive Chemistry XI Dr. B. Kapila,S. K. Khanna,2010-11 Comprehensive chemistry according to the new syllabus prescribed by Central Board of Secondary Education CBSE **Heat Capacity and Thermal Expansion at Low Temperatures** T.H.K. Barron,G.K. White,2012-12-06 The birth of this monograph is partly due to the persistent efforts of the General Editor Dr Klaus Timmerhaus to persuade the authors that they encapsulate their forty or fifty years of struggle with the thermal properties of materials into a book before they either expired or became totally senile We recognize his wisdom in wanting a monograph which includes the closely linked properties of heat capacity and thermal expansion to which we have added a little cement in the form of elastic moduli There seems to be a dearth of practitioners in these areas particularly among physics postgraduate students sometimes temporarily alleviated when a new generation of exciting materials are found be they heavy fermion compounds high temperature superconductors or fullerenes And yet the needs of the space industry telecommunications energy conservation astronomy medical imaging etc place demands for more data and understanding of these properties for all classes of materials metals polymers glasses ceramics and mixtures thereof There have been many useful books including Specific Heats at Low Temperatures by E S Raja Gopal 1966 in this Plenum Cryogenic Monograph Series but few if any that covered these related topics in one book in a fashion designed to help the cryogenic engineer and cryophysicist We hope that the introductory chapter will widen the horizons of many without a solid state background but with a general interest in physics and materials

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