

Periodic Table Basics

Step 1: Complete the square for each element by filling in the atomic number, name, & atomic mass.

Step 2: Determine the number of protons, neutrons, and electrons in an atom of each element.

Step 3: Identify if the element is a solid, liquid, or gas at room temperature.

Step 4: Give the melting (M.P.) and boiling points (B.P.) in degrees Celsius.

Step 5: List at least three physical or chemical properties for each element.

Step 6: List at least three uses for each element.

Step 7: Draw a Bohr diagram and Lewis Structure to show the arrangement of electrons and the number of valence electrons.

 A square card with a large 'B' in the center. To the left of 'B' is a small circle with a plus sign. Below 'B' are two horizontal lines for text.	P = N = E =	O S O L O G
	M.P. = B.P. =	
Properties		
Uses		
Bohr Diagram Lewis Structure A Bohr diagram with two concentric circles and a central circle with a plus sign. To the right is a Lewis structure with a central 'B' and a small circle with a plus sign.		

Step 8: Use the following colors to shade in the square for each element. You should ONLY color the small square in the upper left-hand corner and not the entire card.

Green = Li & Na
Orange = B & Al

Pink = O & S
Red = C & Si

Blue = Be & Mg
Tan = N & P

Purple = F & Cl
Yellow = He, Ne, & Ar

Step 9: Cut the cards apart and arrange according to atomic number in the pattern shown. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

1							2
3	4	5	6	7	8	9	10
11	12	13	14	15	16	17	18

Done? Answer the questions on the worksheet using the information on your Periodic Table!

Heat Calculations Physical Science If8767 Answers

Thomas Preston



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The New Heat Theorem Walther Nernst,1926 Calculations in Heat, Light and Sound ... S. M. Niven,1964

Analysis of Heat Equations on Domains El Maati Ouhabaz,2005 **The Theory of Heat** Thomas Preston,1919

Formulas and tables for heating, German formulas and tables John Henry Kinealy,1899 *New Calculations of the Specific Heat, Enthalpy and Entropy of Air* E. Justi,1937 The Theory of Heat Thomas Preston,1904 **Heat Energy Calculation** ,2006 **Heat, Thermodynamics and Radiation** Alauddin Khan,2021-02-01 Document from the year 2020 in the subject Physics Thermodynamics grade 4 00 language English abstract The book consists of thirteen chapters to fulfill requirements of different kind of readers This volume takes into account the study of Thermometry Kinetic theory of gases the equation of state The change of state Transmission of heat First law of Thermodynamics Thermodynamic functions Second law of Thermodynamics Third law of Thermodynamics Maxwell s equation Clausius Clapeyron equation and Radiation Laws The volume contains illustrative examples of both the ideas and the methods The book is intended as a text book on Heat Thermodynamics and Radiation for undergraduate levels and also as a reference book for anyone who is interested in this field of enquiry The book is comprehensive enough to cover all the topics that are usually taught to upper undergraduate students of Physics Chemistry and Engineering This book will be useful to students and teachers in different universities around the world *Heat Energy Calculation* ,2006 The One-Dimensional Heat Equation John Rozier Cannon,2014-02-19 This is a version of Gevrey s classical treatise on the heat equations Included in this volume are discussions of initial and or boundary value problems numerical methods free boundary problems and parameter determination problems The material is presented as a monograph and or information source book After the first six chapters of standard classical material each chapter is written as a self contained unit except for an occasional reference to elementary definitions theorems and lemmas in previous chapters **Heat Energy Calculation** ,2006 **The Physics of Heat Capacity** B. F. Hann,1971 **Analysis of Heat Equations on Domains** El Maati Ouhabaz,2005 Calculation of Equation for Specific Heat and Specific Heat Ratio as a Function of Temperature Range Limit 600°R to 1800°R ,1964 **On a Formula for Calculating the Expansion of Liquids by Heat** William John Macquorn Rankine,1849

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