

Graphing Periodic Trends Activity

Name: _____ Period: _____

Use the information in these tables to complete the graph as described below

Symbol	Atomic Radius (Picometers)	First Ionization energy (kJoules/mole)	Electronegativity (4-point scale)
H	31	1312	2.1
Li	128	520	1.0
Na	166	496	0.9
K	203	410	0.8
Rb	220	403	0.8
Cs	244	376	0.7

Symbol	Atomic Radius (Picometers)	First Ionization energy (kJoules/mole)	Electronegativity (4-point scale)
Na	166	496	0.9
Mg	141	738	1.2
Al	121	578	1.5
Si	111	787	1.8
P	107	1012	2.1
S	105	1000	2.5
Cl	102	1251	3.0
Ar	106	1521	---

1. Using colored pencils or pen, list under the symbol (in this order!) the Atomic Radius, First Ionization Energy, and Electronegativity.
2. Use a different color for each property. Example: Write all of the Atomic radius values in red, all of the First ionization energies in green, and all of the Electronegativities in blue. Of course, you can pick any colors that are available, as long as you are consistent.
3. Observe the trends in each property as you go down the Alkali metal group, and as you go across Period 3.
4. Write out each of the statements written on the opposite side, completing each statement with the observed trend (increase or decrease)

H	<p>Across a period (→) atomic radius tends to _____.</p> <p>Across a period (→) first ionization energy tends to _____.</p> <p>Across a period (→) electronegativity tends to _____.</p>						
Li							
Na 166 496 0.9	Mg	Al	Si	P	S	Cl	Ar
K	<p>Down a Group (↓) atomic radius tends to _____.</p> <p>Down a Group (↓) first ionization energy tends to _____.</p> <p>Down a Group (↓) electronegativity tends to _____.</p>						
Rb							
Cs							

Laboratory Exercise On Graphing Periodic Trends

Answers

WJ Hussar



Laboratory Exercise On Graphing Periodic Trends Answers:

Resources in Education ,1995 Exploring Physical Science in the Laboratory John T. Salinas ,2019-02-01 This full color manual is designed to satisfy the content needs of either a one or two semester introduction to physical science course populated by nonmajors It provides students with the opportunity to explore and make sense of the world around them to develop their skills and knowledge and to learn to think like scientists The material is written in an accessible way providing clearly written procedures a wide variety of exercises from which instructors can choose and real world examples that keep the content engaging Exploring Physical Science in the Laboratory guides students through the mysteries of the observable world and helps them develop a clear understanding of challenging concepts **The Laboratory Computer** John Dempster,2001-07-10 The Laboratory Computer A Practical Guide for Physiologists and Neuroscientists introduces the reader to both the basic principles and the actual practice of recording physiological signals using the computer It describes the basic operation of the computer the types of transducers used to measure physical quantities such as temperature and pressure how these signals are amplified and converted into digital form and the mathematical analysis techniques that can then be applied It is aimed at the physiologist or neuroscientist using modern computer data acquisition systems in the laboratory providing both an understanding of how such systems work and a guide to their purchase and implementation The key facts and concepts that are vital for the effective use of computer data acquisition systems A unique overview of the commonly available laboratory hardware and software including both commercial and free software A practical guide to designing one s own or choosing commercial data acquisition hardware and software *Chemistry for the IB Diploma Workbook with CD-ROM* Jacqueline Paris,2017-04-06 Chemistry for the IB Diploma Second edition covers in full the requirements of the IB syllabus for Chemistry for first examination in 2016 This workbook is specifically for the IB Chemistry syllabus for examination from 2016 The Chemistry for the IB Diploma Workbook contains straightforward chapters that build learning in a gradual way first outlining key terms and then providing students with plenty of practice questions to apply their knowledge Each chapter concludes with exam style questions This structured approach reinforces learning and actively builds students confidence using key scientific skills handling data evaluating information and problem solving This helps empower students to become confident and independent learners Answers to all of the questions are on the CD ROM

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