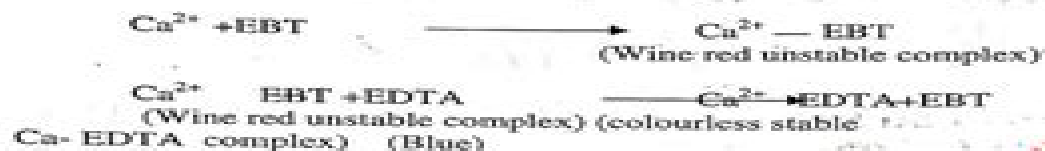
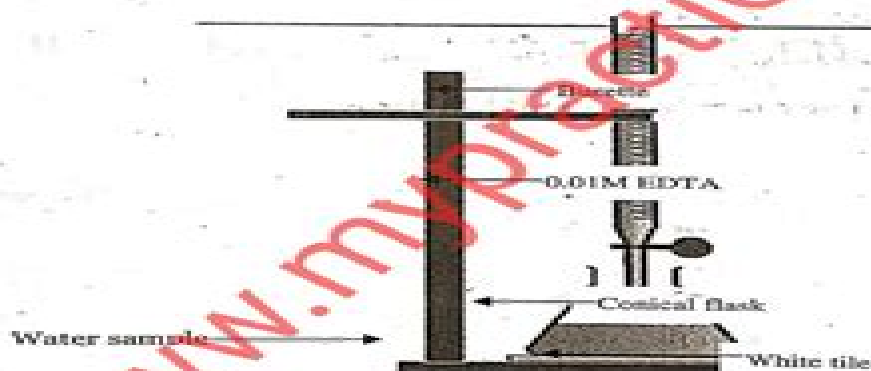


alkaline earth metal ions Ca^{2+} and Mg^{2+} . The EDTA reagent can be used to measure the total quantity of dissolved Ca^{2+} and Mg^{2+} ions in a water sample. Determination of hardness of water is based on the fact that when indicator, Eriochrome Black -T (EBT) is added to hard water in alkaline medium, it forms wine red colour complex with Ca^{2+} and Mg^{2+} ions.



Thus the total hardness of a water sample can be estimated by titration with standard solution of EDTA.

VIII Practical set-up / Circuit diagram / Work Situation

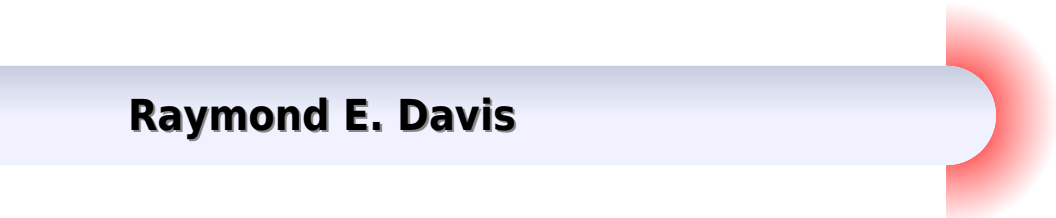


IX Resources Required

S. No.	Name of Resource	Suggested Broad Specification	Quantity
1.	Burette	Borosil glass , Capacity 25 ml /50 ml	One per group
2.	Conical flask	Borosil glass, Capacity 250 ml/100 ml	One per group
3.	Beaker	Borosil glass, Capacity 250 ml/100 ml	One per group
4.	Pipette	Borosil glass, Capacity 25ml / 10 ml,	One per group
5.	EDTA solution	0.01 M	One per group
6.	Buffer solution	pH = 10	As per Requirement
7.	Eriochrome Black T indicator		
8.	Water sample	Hard water sample	

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Table of Contents Msbte Applied Chemistry Answer Key

1. Understanding the eBook Msbte Applied Chemistry Answer Key
 - The Rise of Digital Reading Msbte Applied Chemistry Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Msbte Applied Chemistry Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Msbte Applied Chemistry Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Msbte Applied Chemistry Answer Key

- Personalized Recommendations
 - Msbte Applied Chemistry Answer Key User Reviews and Ratings
 - Msbte Applied Chemistry Answer Key and Bestseller Lists
5. Accessing Msbte Applied Chemistry Answer Key Free and Paid eBooks
- Msbte Applied Chemistry Answer Key Public Domain eBooks
 - Msbte Applied Chemistry Answer Key eBook Subscription Services
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6. Navigating Msbte Applied Chemistry Answer Key eBook Formats
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7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Msbte Applied Chemistry Answer Key
 - Highlighting and Note-Taking Msbte Applied Chemistry Answer Key
 - Interactive Elements Msbte Applied Chemistry Answer Key
8. Staying Engaged with Msbte Applied Chemistry Answer Key
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Msbte Applied Chemistry Answer Key
9. Balancing eBooks and Physical Books Msbte Applied Chemistry Answer Key
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Msbte Applied Chemistry Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Msbte Applied Chemistry Answer Key
- Setting Reading Goals Msbte Applied Chemistry Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Msbte Applied Chemistry Answer Key

- Fact-Checking eBook Content of Msbte Applied Chemistry Answer Key
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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