

AICE Marine Science AS



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Chapter 1 Notes – Quiz 1.2 Solubility in water

Answer Key /25

1. Seawater is a mixture of solutes and solvents. Seawater should be considered a homogeneous solution.

2a) Define the term solute.
A substance that dissolves in a solvent

2b) Describe how sodium chloride dissolves in water.
Through dissolution of ions
Positively charged hydrogen of water molecules surrounds the negative chloride ions
Negatively charged oxygen and hydrogen of water molecules surround the positive sodium ion and the same is done for the other ions

3a) Solubility of sodium chloride can be shown.
 Temperature impacts the solubility of sodium chloride.
As water heats up, the amount of ions found in sodium chloride increases in relation to the solution until it gets saturated.

3b) The average salinity of seawater is 35 ppt. Describe how this body of water is formed.
Through evaporation, which removes water
Will leave g of saltwater until all water evaporates and then solid

3c) Explain how evaporation would impact salinity.
Salinity will rise
As seawater evaporates fresh water is lost from seawater and water is left behind making an increased in salinity

2. The diagram below illustrates changes in dissolved oxygen levels from surface to the bottom of the water.

3a) State the name of the area of extremely low oxygen.
Hypoxia

3b) Describe the trends in dissolved oxygen levels, as seen in the diagram.
As water depth increases, the dissolved oxygen level decreases. The oxygen level is highest at the surface and lowest at the bottom.

3c) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
At the surface, oxygen is produced by photosynthesis. As depth increases, the amount of light decreases, and the amount of oxygen produced decreases. At the bottom, the amount of oxygen is low, and the amount of oxygen is consumed by respiration.

3d) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
At the surface, oxygen is produced by photosynthesis. As depth increases, the amount of light decreases, and the amount of oxygen produced decreases. At the bottom, the amount of oxygen is low, and the amount of oxygen is consumed by respiration.

3e) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
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3f) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
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3g) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
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3h) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
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3i) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
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3k) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
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3p) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
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3q) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
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3r) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
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At the surface, oxygen is produced by photosynthesis. As depth increases, the amount of light decreases, and the amount of oxygen produced decreases. At the bottom, the amount of oxygen is low, and the amount of oxygen is consumed by respiration.

3w) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
At the surface, oxygen is produced by photosynthesis. As depth increases, the amount of light decreases, and the amount of oxygen produced decreases. At the bottom, the amount of oxygen is low, and the amount of oxygen is consumed by respiration.

3x) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
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3y) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
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3z) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
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Date	1 st ppt measurement	2 nd ppt measurement	3 rd ppt measurement	Average ppt
September 7	8.1			
September 8	8.0	8.2		
September 9	8.1	7.9	8.3	8.1
		7.8	8.0	8.0
			8.1	8.0

3a) State what ppt is.
A measure of the concentration of hydrogen ions in a solution.

3b) Describe the method to measure ppt level in a solution.
Using a pH meter or indicator paper to measure the hydrogen ion concentration.

3c) Explain the change in the levels of dissolved oxygen from 0 to 10 m in the diagram.
At the surface, oxygen is produced by photosynthesis. As depth increases, the amount of light decreases, and the amount of oxygen produced decreases. At the bottom, the amount of oxygen is low, and the amount of oxygen is consumed by respiration.

3d) Calculate the average ppt level for September 8.
Average ppt = 8.2 + 7.9 + 8.3 = 8.1

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SJ Ball



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Table of Contents Marine Science Chapter 1review Answers Key

1. Understanding the eBook Marine Science Chapter 1review Answers Key
 - The Rise of Digital Reading Marine Science Chapter 1review Answers Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Marine Science Chapter 1review Answers Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Marine Science Chapter 1review Answers Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Marine Science Chapter 1review Answers Key
 - Personalized Recommendations
 - Marine Science Chapter 1review Answers Key User Reviews and Ratings
 - Marine Science Chapter 1review Answers Key and Bestseller Lists
5. Accessing Marine Science Chapter 1review Answers Key Free and Paid eBooks
 - Marine Science Chapter 1review Answers Key Public Domain eBooks
 - Marine Science Chapter 1review Answers Key eBook Subscription Services
 - Marine Science Chapter 1review Answers Key Budget-Friendly Options

6. Navigating Marine Science Chapter 1review Answers Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Marine Science Chapter 1review Answers Key Compatibility with Devices
 - Marine Science Chapter 1review Answers Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Marine Science Chapter 1review Answers Key
 - Highlighting and Note-Taking Marine Science Chapter 1review Answers Key
 - Interactive Elements Marine Science Chapter 1review Answers Key
8. Staying Engaged with Marine Science Chapter 1review Answers Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Marine Science Chapter 1review Answers Key
9. Balancing eBooks and Physical Books Marine Science Chapter 1review Answers Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Marine Science Chapter 1review Answers Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Marine Science Chapter 1review Answers Key
 - Setting Reading Goals Marine Science Chapter 1review Answers Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Marine Science Chapter 1review Answers Key
 - Fact-Checking eBook Content of Marine Science Chapter 1review Answers Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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