

Objective: Students will identify and differentiate between the different parts and models of the atom.

1. Atoms are made of three basic particles: \_\_\_\_\_
2. What is electric charge? \_\_\_\_\_
3. A positive and a negative charge will \_\_\_\_\_
4. Two positive charges will \_\_\_\_\_
5. Two negative charges will \_\_\_\_\_
6. What does the formula represent? \_\_\_\_\_
7. Electric charge only appears in \_\_\_\_\_ of the elementary charge.
8. What did J.J. Thompson discover? \_\_\_\_\_
9. Atoms have \_\_\_\_\_ charge.
10. The model thought electrons were sprinkled around inside atoms like \_\_\_\_\_ in a \_\_\_\_\_ of \_\_\_\_\_.
11. This was the \_\_\_\_\_ model for the \_\_\_\_\_ of an \_\_\_\_\_, but it wasn't the \_\_\_\_\_ model.
12. In 1911 Rutherford, Geiger and Marsden did an experiment where they shot helium at a thin gold foil. They expected most of the helium to be deflected a little. What actually happened? \_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_
13. The best way to explain this was if a gold atom was mostly \_\_\_\_\_.
14. How much of an atom's mass is contained in the nucleus? \_\_\_\_\_
15. Where is the electron found? \_\_\_\_\_
16. Where is the proton found? \_\_\_\_\_
17. Where is the neutron found? \_\_\_\_\_
18. How much bigger is the electron cloud than the nucleus? \_\_\_\_\_
19. If the atom were a football stadium, how big would the nucleus be? \_\_\_\_\_  
How big would the electron be? \_\_\_\_\_
20. What is the charge of a proton? \_\_\_\_\_
21. What is the charge of an electron? \_\_\_\_\_
22. What is the charge of a neutron? \_\_\_\_\_
23. Why doesn't an electron fall into the nucleus? \_\_\_\_\_
24. What holds the nucleus together? \_\_\_\_\_
25. What are the four types of forces inside an atom? \_\_\_\_\_

# Foundations Of Physical Science Answers To Chapter 9

**L Towne**



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