

Test: Tuesday 2/18/20

*This guide is not the only thing you should use to study. It does not provide you with everything you need. You should also rely on your **Chemical Interactions** textbook, homework, and classroom notes. Use everything you can for the best results.*

*****Study your Solubility and Solutions Quiz***

Topic 1: Understanding Solutions – Textbook pg. 84-89; Binder pg 6-8

1. *What happens to the particles of a solid when dissolved in water? What are the parts of that solution?*

- *Particles of the solid leave each other and become surrounded by particles of water.*
- *Ionic solids in water can conduct electricity; Molecular solids cannot!*
- *Parts of the solution: Solvent:*
 - *Solvent = the largest amount; does the dissolving*
 - *Solute = the smaller amount; gets dissolved*

2. *Compare and contrast solutions, colloids, and suspensions.*

3. *Give an example of each (solutions, colloids, suspensions).*

- *Solutions = Air; saltwater*
- *Colloids = Fog; milk; Gelatin*
- *Solutions = Oil and Vinegar; Pepper/Water*

4. *How do solutes affect a solvent when added? Explain. (3 ways)*

- a. *Lower Freezing Point – Makes it harder for water molecules to form crystals of solid ice*
- b. *Increase Boiling Point – Makes it harder for water molecules to escape*
- c. *Ionic Solutes are able to conduct electricity in water – negative and positive ions are attracted to water molecules*

Contrast		
Solutions	Colloids	Suspensions
<i>*Same properties throughout *Molecules are too small to be seen **Can be made from solids, liquids, or gases</i>	<i>*Larger particles than a solution but still too small to be seen. **Able to scatter a beam of light</i>	<i>Particles can be seen and easily separated by settling or filtration DOES NOT have the same properties throughout</i>

Topic 2: Concentration and Solubility – Textbook pg. 92-97; Binder pg 9-12

5. *What is the difference between a saturated and an unsaturated solution? Give examples.*

- *Saturated - add so much solute that no more dissolves*
- *Unsaturated – Able to dissolve more solute*

6. *Explain the difference between a dilute and a concentrated solution. How is this different from a saturated and unsaturated solution?*

- *Dilute - Mixture that has only a little solute dissolved in a solvent*
- *Concentrated – Has a lot of solute dissolved in the solvent*
- *Saturated and Unsaturated solutions compares the ability of a solute to dissolve.*

Ms Study Guide Answer Key

**Kevin T. Patton, Frank B. Bell, Terry
Thompson, Peggie L. Williamson, Eric L
Sun**

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Study Guide for The Human Body in Health & Disease - E-Book Kevin T. Patton, Frank B. Bell, Terry Thompson, Pegg L. Williamson, Eric L. Sun, 2023-03-02 Reinforce your understanding of the concepts in Patton's The Human Body in Health Disease. Corresponding to the chapters in the text, this study guide reviews essential medical terminology concepts and processes related to anatomy and physiology and explains how body systems function in health and disease. Each chapter begins with a quick synopsis of the key points in the textbook chapter. A variety of exercises make it easier to review and apply key concepts, and labeling of anatomy drawings helps in learning anatomical terms and structures. Know your Medical Terms feature helps familiarize you with the various word parts used in medical terminology and pairs nicely with the Language of Medicine word lists in the main text to reinforce medical terminology concepts that are key to understanding. A P Brief synopsis of core textbook concepts provides a comprehensive review of essential content. Crossword Puzzles and Word Finds help you master new vocabulary terms. Application Questions ask you to make judgments based on the information in the chapter. Diagrams and labeling exercises help reinforce where the structures of the body are located. Matching and fill in the blank exercises aid in understanding chapter content. Answers to exercises are provided in the back of the guide along with references to the appropriate textbook page and the textbook objective so that you can easily find where the material is presented. Study tips in the Preface provide an overview of the most effective methods for learning and retaining information.

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