

Name: _____

Date: _____

Mark: _____

Experiment No. 05: Effect of Heat on Soluble

Substance (Sulphuric acid) (Experiment No. 05.01)

The student's observations and the resulting chemical reactions. The most important part is the number of precipitates formed in each case. The number of precipitates formed in each case is the number of precipitates formed in each case. The number of precipitates formed in each case is the number of precipitates formed in each case. The number of precipitates formed in each case is the number of precipitates formed in each case.

The student has to determine the chemical reactions in this experiment. The student has to determine the chemical reactions in this experiment. The student has to determine the chemical reactions in this experiment. The student has to determine the chemical reactions in this experiment.

Objectives:

- 1. To determine the number of moles of water produced in the reaction of sulphuric acid with sodium hydroxide.
- 2. To determine the number of moles of water produced in the reaction of sulphuric acid with sodium hydroxide.
- 3. To determine the number of moles of water produced in the reaction of sulphuric acid with sodium hydroxide.
- 4. To determine the number of moles of water produced in the reaction of sulphuric acid with sodium hydroxide.

Pre-lab Questions:

1. The student has to determine the number of moles of water produced in the reaction of sulphuric acid with sodium hydroxide. The student has to determine the number of moles of water produced in the reaction of sulphuric acid with sodium hydroxide. The student has to determine the number of moles of water produced in the reaction of sulphuric acid with sodium hydroxide.

2. The student has to determine the number of moles of water produced in the reaction of sulphuric acid with sodium hydroxide. The student has to determine the number of moles of water produced in the reaction of sulphuric acid with sodium hydroxide. The student has to determine the number of moles of water produced in the reaction of sulphuric acid with sodium hydroxide.

3. The student has to determine the number of moles of water produced in the reaction of sulphuric acid with sodium hydroxide. The student has to determine the number of moles of water produced in the reaction of sulphuric acid with sodium hydroxide. The student has to determine the number of moles of water produced in the reaction of sulphuric acid with sodium hydroxide.

Table 1: Results of the experiment

Before the reaction:	Mass of sulphuric acid: _____
	Mass of sodium hydroxide: _____
	Mass of water: _____
After the reaction:	Mass of sulphuric acid: _____
	Mass of sodium hydroxide: _____

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Padhraic Smyth



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