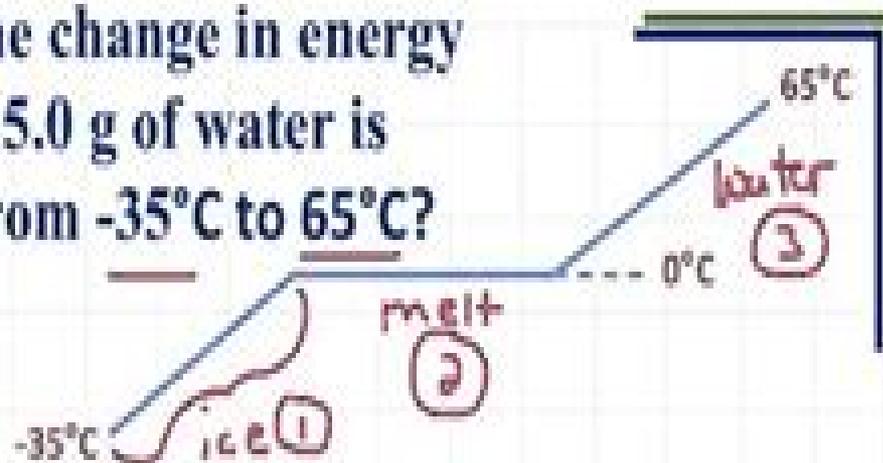


What is the change in energy
when 25.0 g of water is
heated from -35°C to 65°C ?



$$\textcircled{1} \quad q = m C_s \Delta T = 25.00\text{g} \left(\frac{2.09\text{J}}{\text{g}^{\circ}\text{C}} \right) (0 - (-35))$$
$$q = 1829\text{J}$$

$$\textcircled{2} \quad q = n \Delta H_{\text{fus}} = 25.0\text{g} \left(\frac{\text{mol}}{18.01\text{g}} \right) \left(\frac{6.02\text{kJ}}{\text{mol}} \right) (1)$$

Heating Curve Problems Key

CL Gary



Heating Curve Problems Key:

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In a world driven by information and connectivity, the power of words has become much more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such is the essence of the book **Heating Curve Problems Key**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

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