Name:

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## BALANCING AND IDENTIFYING CHEMICAL EQUATIONS

Classify the following reactions and balance them.

Al<sub>2</sub>(SO<sub>2</sub>)<sub>3</sub> + BaCl<sub>2</sub> → BaSO<sub>4</sub> + AlCl<sub>2</sub>

Type of reaction:

(2) Al,S, → Al + S

Type of reaction:

(3) NaOH + CuSO, → Na,PO, + Cu(OH), Type of reaction:

(4) Fe + H,SO, → Fe,SO, + H,

Type of reaction:

(5) C<sub>4</sub>H<sub>12</sub> + O<sub>2</sub> → CO<sub>2</sub> + H<sub>2</sub>O

Type of reaction:

(6) H,S + O, → SO, + H,O

Type of reaction:

(7) C,H,O + O, → CO, + H,O

Type of reaction:

(8) Al + NiBr, → AlBr, + Ni

Type of reaction:

(9) AI + O, → AI,O,

Type of reaction:

(10) H<sub>2</sub>O<sub>2</sub> → H<sub>2</sub>O + O<sub>2</sub>

Type of reaction:

(11) K + CI, → KCI

Type of reaction:

(12) Na + MgCl, -> NaCl + Mg

Type of reaction:

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