SPECIFIC HEAT WORKSHEET

Specific heat is the quantity of heat required to raise the temperature of one gram of a substance by one Celsius degree.



where Q = heat energy, m = mass, and ΔT = change in temp. Remember, ΔT = (Tfinal – Tinitial).

Directions : calculating specific heat

 A 15.75-g piece of iron absorbs 1086.75 joules of heat energy, and its temperature changes from 25°C to 175°C. Calculate the specific heat capacity of iron.

$$C = Q = 1086.75 = 0.46 \text{ J/g}^{\circ}\text{C}$$

 $m(T_f-T_i) = 15.75(175-25)$

2. How many joules of heat are needed to raise the temperature of 10.0 g of aluminum from 22°C to 55°C, if the specific heat of aluminum is 0.90 $J/g^{\circ}C$?

$$Q = mC(T_1 - T_1) = 10.0g (0.90J/g^{\circ}C)(55-22) = 297 J$$

 Calculate the specific heat capacity of a piece of wood if 1500.0 g of the wood absorbs 67,500 joules of heat, and its temperature changes from 32°C to 57°C.

$$C = Q = \frac{67500 \text{ J}}{m(T_f - T_f)} = \frac{1.8 \text{ J/g}^{\circ}\text{C}}{(1500 \text{ g})(57-32)}$$

 100.0 g of 4.0°C water is heated until its temperature is 37°C. Calculate the amount of heat energy needed to cause this rise in temperature.

$$Q = mC(T_{f}-T_{i}) = 100g(4.184J/g^{\circ}C)(37 - 4) = 14000 J$$

25.0 g of mercury is heated from 25°C to 155°C, and absorbs 455 joules of heat in theprocess. Calculate the specific heat capacity of mercury.

$$C = Q = 455 J = 0.14 J/g^{\circ}C$$

 $m(T_f-T_i)$ (25g)(155-25)





Heat And Temperature Student Sheet Answers

Ann Lawrence, Charlie Hennessy

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Page, Tristram Shepard, 1999 Designed to help pupils develop their IT capability and apply ICT across all subject areas in line with National Curriculum 2000 requirements This work offers pupil focused material which allows independent learning and progress at individual rates
Hands-On Science for Manitoba, Grade 7 Jennifer Lawson, 2004 Hands On Science for Manitoba is filled with a year s worth of classroom tested activities The grade seven book is custom written to the Manitoba science curriculum 2000 and divided into four units Interactions Within Ecosystems Particle Theory of Matter Forces and Structures Earth s Crust This teacher resource offers a detailed introduction to the Hands On Science program which includes its guiding principles implementation guidelines an overview of the science skills that grade 7 students use and develop a classroom assessment plan complete with record keeping templates and digital reproducible student materials

ICONECT 2019 Murtono, Moh Kanzunnudin, Fina Fakhriyah , Ratri Rahayu, Robbi Rahim, The complex problems of education and technological development and information demands then takes its main innovations in learning The purpose of this Education is Innovation in order to improve the quality effectiveness efficiency relevance and productivity making the learning process more meaningful and fun for children Innovation can be performed in all subjects learning methods media and evaluation Innovation based learning local culture values will yield the superior character that will benefit children in the face of a globalized world So is innovation technology based learning make learning be fun so that children become active and creative ideas thoughts research related to the innovation of education can be presented in International Conference Education Culture and technology is preferred The theme of this Conference Innovation of Education to Improve Character **Properties of Matter: Chemical Changes and Chemical Properties Gr. 5-8** George Value for Childern Graybill, 2015-09-01 This is the chapter slice Chemical Changes and Chemical Properties from the full lesson plan Properties of Matter Discover what matter is and is not Learn about and the difference between a mixture and a solution Chocked full with hands on activities to understand the various physical and chemical changes to matter Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Written to grade these science concepts are presented in a way that makes them more accessible to students and easier to understand Our resource is jam packed with experiments reading passages and activities all for students in grades 5 to 8 Color mini posters and answer key included and can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom's Taxonomy and STEM initiatives E3 Chemistry AP Exam Practice - 2018: With Answers, Explanations and Scoring Guidelines Effiong Eyo, 2018-01-15 Preparing for Chemistry AP Exam has never been easier more enticing more exciting more engaging more understandable and less overwhelming Our book is written to help students do more know more and build confidence for a higher mark on their AP exam With a total of four practice tests with answers

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