Answer: D. 5.0

7. Calculate the concentration of hydrogen ions in the following acetic acid solution. pK<sub>a</sub> = 4.76 [CH<sub>3</sub>COOH] / [CH<sub>3</sub>COO<sup>-</sup>] = 0.5

A. 7.89\*10-5M

# **Gre Chemistry Guide**

**Christian Drosten** 

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