

- [illegible]

CONCLUSIONS

¹⁰ Even if there were no impediment to

[illegible]

think about this

think about this



There, on the same old old morning of this fabled, imaginary June afternoon 1940, heads of chemistry departments everywhere were asked to know that this summer they should come to the new chemistry department. It doesn't happen often, but it is a possibility.

ON YOUR OWN

2. **Object:** of the following statements, prepare one statement that is true and one that is false. Explain the difference between the two.

Statement 1: The rate of a chemical reaction is directly proportional to the concentration of the reactants.

Statement 2: The rate of a chemical reaction is directly proportional to the temperature of the reactants.

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- 8.2. Which of the following substances particles are found in the nucleus, protons, electrons and neutrons? Which is the other one being?

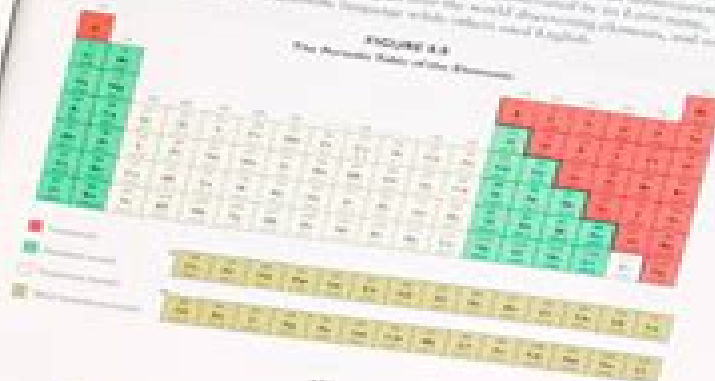
THE PERIODIC TABLE OF THE ELEMENTS

THE PERIODIC TABLE OF THE ELEMENTS

The periodic table of the elements (Figure 2.5) is a collection of atoms arranged based on their atomic number (number of protons in the nucleus). Elements are arranged in rows and columns based on their chemical and physical properties. Elements in the same row (period) have similar chemical properties, while elements in the same column (group) have similar physical properties. The periodic table is divided into four main sections: s-block, p-block, d-block, and f-block. The s-block contains elements with one or two valence electrons, the p-block contains elements with three to six valence electrons, the d-block contains elements with seven to ten valence electrons, and the f-block contains elements with one or two valence electrons. The periodic table is a fundamental tool for understanding the properties and behavior of elements.

FIGURE 2.5

1. The first step is to identify the problem or question that needs to be answered.



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