

GENERAL PSYCHOLOGY FINAL EXAM STUDY GUIDE

1. In its early years, psychology focused on the study of ____, but from the 1920s into the 1960s, American psychologists emphasized the study of ____.
A. environmental influences; hereditary influences
B. maladaptive behavior; adaptive behavior
C. unconscious motives; conscious thoughts and feelings
D. mental life; observable behavior
2. The young science of psychology developed from the more established fields of philosophy and ____.
A. economics
B. biology
C. geography
D. sociology
3. Janna has low self-esteem because she is often teased for being overweight. Appreciating the complexity of Janna's difficulties requires ____.
A. introspection
B. psychoanalysis
C. massed practice
D. a biopsychosocial approach
4. Professor Crisman believes that most women prefer tall and physically strong partners because this preference promoted the survival of our ancestors' genes. This viewpoint best illustrates the ____ perspective.
A. social-cultural
B. cognitive
C. evolutionary
D. psychodynamic
5. Professor Lopez believes that severe depression results primarily from an imbalanced diet and abnormal brain chemistry. Professor Lopez favors a ____ perspective on depression.
A. neuroscience
B. psychodynamic
C. behavior genetics
D. cognitive
6. Efforts to discover whether the intelligence of children is more heavily influenced by their biology or by their home environments are most directly relevant to the debate regarding ____.
A. structuralism vs. functionalism
B. evolution vs. natural selection
C. observation vs. introspection
D. nature vs. nurture
7. Contemporary psychology is best defined as the science of ____.
A. conscious and unconscious mental activity
B. observable responses to the environment
C. behavior and mental processes
D. maladaptive and adaptive behaviors
8. Humanistic psychologists focused attention on the importance of people's

General Psychology Exam 1 Study Guide Answers

Katherine Demitrakis



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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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 $\text{HCl} + \text{NaOH} \rightarrow \text{NaCl} + \text{H}_2\text{O}$
1 mole of HCl would be fully neutralized by one mole of NaOH

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web this point of titration is referred to as the equivalence point considering the 1 1 stoichiometry of this acid base reaction NaOH aq $\text{C}_6\text{H}_4\text{COOH}$ aq $\text{C}_6\text{H}_4\text{COONa}$ aq H_2O l the point of equivalence is the point of titration when the number of moles of naoh na added is equal to the number of moles of khp nb in the

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