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MPUMALANGA PROVINCE
REPUBLIC OF SOUTH AFRICA

**PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 1 FORMAL EXPERIMENT - PART 2
VERIFICATION OF BOYLE'S LAW
MARKING GUIDELINES**

- 1.1 • Aim stated correctly. ✓
e.g. To determine the relationship between the volume of a fixed amount of gas and the pressure on the gas when temperature remains constant.

(1)

- 1.2 • All apparatus listed ✓

1. Boyle's Apparatus
2. Pump

- All steps of the method correct and logically stated. ✓✓ correct tense used. ✓✓

e.g.

1. The tap on the Boyle's Law apparatus was opened and then the pump was connected.
2. The apparatus was pumped to the pressure.
3. The oil was allowed to settle for a few minutes.
4. A table was used to record the volume reading on the tube and the corresponding pressure reading on the Bourdon pressure gauge.
5. The air tap was slightly opened and quickly closed to let a little air escape from the system. The oil was allowed to settle for a few minutes and then the pressure and volume readings were taken.
6. 5 different readings were recorded in the table.

(5)

2. Observations and Recording of results:

Learners will use results obtained during experiment to complete the table below.

(6)

Volume (V) cm ³	Pressure (p) kPa		pV (✓)

Marking guidelines:

- ✓ for correct values of the volume
- ✓ for correct values of the pressure
- ✓✓ for correct values of

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Table of Contents Grade 11 Memo Practical Test October 2014

1. Understanding the eBook Grade 11 Memo Practical Test October 2014
 - The Rise of Digital Reading Grade 11 Memo Practical Test October 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade 11 Memo Practical Test October 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 11 Memo Practical Test October 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 11 Memo Practical Test October 2014
 - Personalized Recommendations
 - Grade 11 Memo Practical Test October 2014 User Reviews and Ratings
 - Grade 11 Memo Practical Test October 2014 and Bestseller Lists
5. Accessing Grade 11 Memo Practical Test October 2014 Free and Paid eBooks
 - Grade 11 Memo Practical Test October 2014 Public Domain eBooks
 - Grade 11 Memo Practical Test October 2014 eBook Subscription Services
 - Grade 11 Memo Practical Test October 2014 Budget-Friendly Options
6. Navigating Grade 11 Memo Practical Test October 2014 eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade 11 Memo Practical Test October 2014 Compatibility with Devices
 - Grade 11 Memo Practical Test October 2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade 11 Memo Practical Test October 2014
 - Highlighting and Note-Taking Grade 11 Memo Practical Test October 2014
 - Interactive Elements Grade 11 Memo Practical Test October 2014
8. Staying Engaged with Grade 11 Memo Practical Test October 2014

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Grade 11 Memo Practical Test October 2014
- 9. Balancing eBooks and Physical Books Grade 11 Memo Practical Test October 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 11 Memo Practical Test October 2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade 11 Memo Practical Test October 2014
 - Setting Reading Goals Grade 11 Memo Practical Test October 2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade 11 Memo Practical Test October 2014
 - Fact-Checking eBook Content of Grade 11 Memo Practical Test October 2014
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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