

AP CHEMISTRY UNIT 8: EQUILIBRIUM CHANGES: Le Chatelier's Principle

Unit 8 - Introduction to Equilibrium

1. Which of the following is NOT true?

Chemical equilibrium is a dynamic equilibrium where both the forward and reverse reactions are occurring at the same rate.

At equilibrium, the concentrations of reactants and products are equal.

At equilibrium, the rates of the forward and reverse reactions are equal.

At equilibrium, the system is at a minimum of free energy.

At equilibrium, the system is at a maximum of free energy.

At equilibrium, the system is at a minimum of free energy.

☒ At equilibrium, the system is at a maximum of free energy.

2. Which of the following is NOT a characteristic of a system at equilibrium?

The concentrations of reactants and products are constant.

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The system is at a minimum of free energy.

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At equilibrium, the system is at a minimum of free energy.

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5. Which of the following is NOT a characteristic of a system at equilibrium?

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6. Which of the following is NOT a characteristic of a system at equilibrium?

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Equilibrium Ap Test Multiple Choice

Karin Nielsen-Saines



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