

1st Semester Chemistry Final Exam Review

1. Identify the following reaction type AND correct products:



- a. combustion
- b. synthesis
- c. decomposition
- d. single replacement
- e. double replacement

- w. $\text{Ag} + \text{ScOH}$
- x. $\text{Ag} + 2 \text{Sc}(\text{OH})_2$
- y. $3 \text{Ag} + \text{Sc}(\text{OH})_3$
- z. $3 \text{Ag} + 2 \text{Sc}(\text{OH})_4$

- | | |
|-------|---------|
| I. | a. & w. |
| II. | b. & z. |
| III. | b. & y. |
| IV. | d. & z. |
| V. | e. & z. |
| VI. | d. & y. |
| VII. | e. & y. |
| VIII. | c. & z. |
| IX. | d. & x. |

2. Convert: $2.32 \times 10^5 \text{ mg} = ? \text{ kg}$

3. Calculate the energy for a quantum of light with a frequency of $8.44 \times 10^{16} \text{ Hz}$.

4. Determine the oxidation number for iodine in aluminum iodate.

5. REPORTING WITH THE CORRECT NUMBER OF SIGNIFICANT DIGITS, how many grams are in **4.011 moles** of calcium chloride?

- a) 445.14078 grams b) 445.140 grams c) 445.1 grams d) 445 grams
e) 450 grams f) 400 grams

6. Calculate the frequency of light that has a wavelength of 210 nm.

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Katrin Zwirgmaier



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Table of Contents Honors Chemistry Practice Test

1. Understanding the eBook Honors Chemistry Practice Test
 - The Rise of Digital Reading Honors Chemistry Practice Test
 - Advantages of eBooks Over Traditional Books
2. Identifying Honors Chemistry Practice Test
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Honors Chemistry Practice Test
 - User-Friendly Interface
4. Exploring eBook Recommendations from Honors Chemistry Practice Test
 - Personalized Recommendations
 - Honors Chemistry Practice Test User Reviews and Ratings
 - Honors Chemistry Practice Test and Bestseller Lists
5. Accessing Honors Chemistry Practice Test Free and Paid eBooks
 - Honors Chemistry Practice Test Public Domain eBooks
 - Honors Chemistry Practice Test eBook Subscription Services

- Honors Chemistry Practice Test Budget-Friendly Options
- 6. Navigating Honors Chemistry Practice Test eBook Formats
 - ePub, PDF, MOBI, and More
 - Honors Chemistry Practice Test Compatibility with Devices
 - Honors Chemistry Practice Test Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Honors Chemistry Practice Test
 - Highlighting and Note-Taking Honors Chemistry Practice Test
 - Interactive Elements Honors Chemistry Practice Test
- 8. Staying Engaged with Honors Chemistry Practice Test
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Honors Chemistry Practice Test
- 9. Balancing eBooks and Physical Books Honors Chemistry Practice Test
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Honors Chemistry Practice Test
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Honors Chemistry Practice Test
 - Setting Reading Goals Honors Chemistry Practice Test
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Honors Chemistry Practice Test
 - Fact-Checking eBook Content of Honors Chemistry Practice Test
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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